Improved Photovoltaic Response of a Near-Infrared Sensitive Solar Cell by a Morphology-Controlling Seed Layer

Lushuai Zhang,[†] Trisha L. Andrew^{†§}*

^[†] Department of Materials Science and Engineering, University of Wisconsin-Madison,
Madison, Wisconsin 53706

^[§] Department of Chemistry, University of Wisconsin-Madison, Madison, Wisconsin 53706

^[*] tandrew@chem.wisc.edu

Abstract

We demonstrate that a thin seed layer of indium phthalocyanine chloride (ClInPc) annealed under mild conditions effectively controls the morphology of both post-annealing deposited ClInPc films and ClInPc:C₆₀ mixed films, introducing the triclinic phase into the commonly monoclinic phase dominating film. ClInPc/C₆₀ planar solar cells and ClInPc:C₆₀ (1:1) planar-mixed solar cells with and without the triclinic phase were studied. Increased short circuit current (J_{sc}), fill factor (FF), external quantum efficiency (EQE) and internal quantum efficiency (IQE) of the devices containing triclinic phase is attributed to the enhanced absorption in the near infrared (NIR) region and decreased series resistance. The correlation between open circuit voltage (V_{oc}) and dark saturation pre-exponential factor (J_{s0}) was analyzed to investigate V_{oc} loss upon annealing. The overall performance of device is considerably improved by introducing the triclinic phase of ClInPc.

Keywords:

Organic solar cells, non-planar Phthalocyanine, molecular aggregate phase

Graphical abstract

1. Introduction

The compatibility with low-cost materials and processing renders organic photovoltaics (OPVs) great potential to contribute to future energy needs¹⁻². However, although the power conversion efficiency of organic solar cells has steadily increased due to tremendous efforts in tuning molecular structure, morphology and device architecture³⁻⁸, the performance of OPVs requires further improvement for commercial application. One of the limiting factors is solar spectrum loss arising from narrow absorption range of each particular conjugated organic small molecule or polymer. One approach to this issue is to establish tandem devices by stacking subcells with photoactivity in different wavelength regions, thus ensuring wide spectrum coverage⁹⁻¹². Most organic solar cells underperform in the near infrared (NIR) wavelength region, which is a region of high solar photon flux, with 40% of the solar power being lost due to inefficient absorption in the NIR. Therefore, optimizing NIR absorption offers the opportunity to further increase the short circuit current of OPVs¹³⁻¹⁶.

Donor materials that have NIR response include donor-acceptor conjugated oligomers and polymers^{9, 17}, carbon nanotubes¹⁸⁻¹⁹, and phthalocyanine (Pc) materials^{15, 20-24}. Various Pc materials are most frequently used due to their chemical tunability. Incorporating metal centers into Pc ring changes the shape of the molecule, which, in turn, tunes the morphology, crystallinity, and optical and electronic properties of the resulting films²⁴. Divalent metals, such as Cu, Zn, Co and Ni, give planar metal-Pc complexes, which form herringbone stacks in films. Two similar polymorphs, assigned as the α - and β - phase, are induced from two different intermolecular tilt angles within the herringbone stack²⁵⁻²⁶. The high similarity of the two phases gives rise to similar absorption features, with the most intense Q-band located between 500 – 750 nm^{21, 27}. On the contrary, non-planar Pc materials, including PbPc, ClAlPc, TiOPc, VOPc and ClInPc, exhibit vastly different polymorphs, which are assigned as monoclinic phase and triclinic phase²⁸⁻³⁰. Monoclinic form consists of molecules stacking in a face-to-face fashion, giving rise to metal ions in a linear chain^{28, 31}. Triclinic form consists of molecules stacking in a head-to-tail fashion, alternately orienting convex and concave sides^{28, 32}. The monoclinic form displays a blue-shifted absorption band compared to the monomer absorption, while the triclinic phase displays a red-shifted absorption band compared to the monomer^{22, 33-36}. In PbPc, for example, the absorption maximum of the triclinic form is at 900 nm while that of the monomer is at 690 nm^{23, 33}. Thus, the triclinic phase of non-planar Pc materials, in particular, renders them very interesting NIR absorbers for organic solar cells. Phase transformations between monoclinic and triclinic and/or mixtures of the two phases in thin films can be accessed by varying the deposition rate, tuning the substrate temperature during deposition, solvent annealing, or templating via a crystalline interlayer.

Tuning the organic film crystallinity, molecular orientation and aggregation by a templating layer toward better performance of organic solar cells has been drawing interest. The templating layers are focused on semiconducting organic molecules³⁷⁻³⁹ and copper halide.^{33, 35, 40-41} However, organic templating materials would cause competent light absorption, as well as additional interfacial energetics to be considered. Copper halide is light sensitive and is detrimental to stability of organic materials.⁴² The aforementioned limitations evoke endeavors to investigate alternative strategies to achieve the same outcome.

In this work, we demonstrate that annealing a thin (5 nm) seed layer of ClInPc under mild conditions is a useful strategy to introduce the triclinic phase in subsequently deposited ClInPc

film or ClInPc:C₆₀ mixed films. Triclinic phase evolution was proved by morphology and absorption studies. Introducing the triclinic phase into ClInPc solar cells enhances device absorption and EQE beyond the 800 nm region and significantly increases IQE throughout the entire ClInPc photoactive region, which leads to increased J_{sc}, FF and power conversion efficiency (PCE) in both planar and planar-mixed solar cells. The loss of V_{oc} with increasing triclinic phase was investigated and was explained by an increased dark saturation pre-exponential factor, J_{so}. The device performance is consistent with the previous study that adjacent molecules in triclinic phase have stronger π - π interaction than those in monoclinic phase³⁴. On one hand, the stronger intermolecular interaction is responsible for the more delocalized electronic states³⁴ favoring charge generation and collection. On the other hand, the stronger intermolecular interaction enhances recombination at donor/acceptor (D/A) interfaces leading to higher J_{so}. Hence, the molecular aggregate structure in non-planar phthalocyanine films needs to be carefully studied and engineered to fully utilize this NIR sensitive solar cell material.

2. Materials and Methods

ClInPc was purchased from Lumtec Corp., MoO₃, C₆₀ and BCP were purchased from Sigma-Aldrich Company. Organic thin films were grown on 10 nm MoO₃ pre-deposited on indium-tin-oxide coated glass substrate (ITO/glass) in high vacuum (<10⁻⁶ Torr) at rate of 0.5 Å/s, with the substrate maintained at room temperature during deposition. ClInPc:C₆₀ (1:1) blend films were prepared by co-deposition with both rates at 0.5 Å/s. Annealing of 5 nm seed layer of ClInPc was conducted on a hot plate in N₂ environment at 60°C, 80°C or 100°C for 15 min. MoO₃ and the subsequent organic layers were deposited without a mask, while the top Ag electrode was deposited with a metal mask to yield a final device area of 1.21 mm².

The optical absorption was obtained by measuring the transmittance and reflectance of the ClInPc films on ITO/glass deposited with 10 nm MoO₃ using an Evolution 220 UV-Visible Spectrophotometer with ISA 220 integrating sphere. The morphology of the films was characterized in air using an Agilent 5500 atomic force microscope (AFM). All devices related operation and measurement were performed in a nitrogen filled glovebox. Current-voltage characteristics of organic solar cells was measured under dark and simulated AM 1.5G solar illumination from a solar simulator with a Xe-arc lamp. A crystalline Si reference cell was used to measure the intensity of the solar simulator, which was adjusted to 1 sun.

3. Results and Discussion

3.1 Film morphology

First, we characterized the morphologies of a 5 nm seed layer of ClInPc as-deposited or annealed at different temperatures (shown in Figure 1). Figure 1a of bare ITO surface morphology shows clusters of tens of nanometers in diameter with root mean square (RMS) roughness of 1.23 nm. A 5 nm ClInPc seed layer (Figure 1b-e) as-deposited or annealed covers the ITO cluster morphology, suggesting the presence of a continuous wetting layer. The as-deposited film is mostly uniform with sparsely distributed islands. RMS roughness of ClInPc films increases along with the annealing temperature, which is 2.2 nm for as-deposited film, 3.2 nm, 5.6 nm and 6.1 nm for films annealed at 60°C, 80°C or 100°C, respectively. The roughness partially arises from the lateral growth of islands. To understand the vertical growth behavior of

islands at different temperatures, the height density distribution was plotted in Figure 1f. The as-deposited ClInPc film exhibits a density maximum at 4.7 nm, a shoulder at 5.8 nm, and a tail extending to 10 nm. Upon annealing, all density distributions are broadened, with the maxima shifting to 5.5 nm. The distribution tail extends to 12 nm for the film annealed at 60°C, and to 30 nm for the film annealed at 80°C or 100°C. Therefore, we can conclude that larger size clusters evolve with higher annealing temperature.

Next, we investigated the morphologies of films with an additional 15 nm ClInPc grown on the aforementioned 5 nm seed layers (Figure 2a-d). Significant differences in film morphology were observed for films grown on seed layers annealed at 80°C and higher. The domain size of ClInPc films also increased with increasing seed layer annealing temperature: RMS roughness of 15 nm ClInPc films grown on seed layers as-deposited or annealed at 60°C, 80°C or 100°C were 3.5 nm, 4.2 nm, 6.7 nm and 7.1 nm respectively.

Figure 2e-f shows surface morphology of ClInPc:C₆₀ (1:1) blend films used for planar-mixed solar cells. The films were prepared as follows: seed layer either as-deposited or annealed at 80°C was prepared on MoO₃ covered ITO/glass. An additional 5 nm ClInPc and 10 nm ClInPc:C₆₀ (1:1) blend was subsequently deposited. The surface of a ClInPc:C₆₀ (1:1) blend film is smoother than that of neat ClInPc films. The RMS roughness of the blend film on the seed layer as-deposited is 1.2 nm and on the annealed seed layer is 5.6 nm. The observed larger cluster size and higher RMS roughness of the ClInPc films and ClInPc:C₆₀ (1:1) blend films grown on the annealed seed layer suggest better crystallinity.

Figure 1. AFM topography images of (a) bare ITO, (b) 5 nm films of ClInPc seed layers as-deposited, (c) annealed at 60°C, (d) 80°C and (e) 100°C. (f) Shows the height distribution of images (b)-(e).

Figure 2. AFM topography images of (a) 15 nm film of ClInPc grown on seed layers as-deposited, (b) 15 nm film of ClInPc grown on seed layers annealed at 60°C, (c) 80°C, (d) and 100°C. (e) ClInPc:C₆₀ (1:1) mixed film grown on 10 nm ClInPc film with first 5 nm as deposited and (f) annealed at 80°C.

3.2 Film absorption

Figure 3a shows the absorption of 15 nm ClInPc film grown on 5 nm ClInPc seed layer. As was observed for non-planar phthalocyanine materials, PbPc, VOPc and TiOPc, the triclinic phase induces stronger π - π interaction between adjacent Pc rings and causes broadening of the Q-band spectrum toward NIR.⁴³⁻⁴⁸ The reason for this broadening is the prevalent exciton coupling in the coordinated complex.^{22, 49} For the as-deposited film, the dominant peak at 730 nm can be assigned to the monoclinic phase of ClInPc. Upon annealing at 60°C, a very weak shoulder evolves at 830 nm, which can be assigned to the triclinic phase of ClInPc. Increasing

absorption at 830 nm, corresponding to an increasing concentration of the triclinic phase, is observed with increasing annealing temperature. Examining the integral of the total absorption spectra of these films, it is apparent that the integral of all 20 nm ClInPc film are identical at wavelengths below 800 nm but diverge above 800 nm. The triclinic phase in films annealed at 80°C or 100°C significantly contribute to the total absorption.

To tease out whether the phase transformation occurs within the 5 nm seed layer or in the subsequently-deposited 15 nm film, the absorption of 5 nm seed layer as-deposited and annealed at 80°C was measured as shown in Figure 3b. For the as-deposited film, there is no shoulder at 830 nm. The ratio of absorption at 830 nm and at 730 nm (A_{830}/A_{730}) is 0.084. For the annealed seed layer, A_{830}/A_{730} is 0.17, indicating triclinic phase formation in the seed layer. However, A_{830}/A_{730} of the film with additional 15 nm film grown on seed layer annealed at 80°C is 0.40, about 2.4 times higher than the value of the seed layer, suggesting that most of the triclinic phase is formed in the upper 15 nm film.

Figure 3c shows the absorption spectra of the same planar-mixed films used for the morphological study in Figure 2e-f. A weak triclinic absorption appears in the bulk film grown on the as-deposited seed layer, with an A_{830}/A_{730} value of 0.095. Films grown on a seed layer annealed at 80°C display an increased A_{830}/A_{730} ratio of 0.27. Thus, the annealed seed layer also directs the formation of triclinic phase in ClInPc:C₆₀ blend films.

Figure 3. (a) Absorption spectra of 15 nm film of ClInPc grown on seed layers as-deposited or annealed and the integral of each spectrum. The measurement was taken under transmittance and reflectance mode respectively. The evolution of absorption above 800 nm of the films with annealed seed layer is attributed to the triclinic phase of ClInPc, which also contributes to the enhancement of total absorption. (b) Absorption spectra of 5 nm ClInPc seed layers as deposited or annealed at 80°C. (c) Absorption spectra of ClInPc:C₆₀ (1:1) mixed film grown on a ClInPc seed layer as-deposited and annealed at 80°C.

3.3 Device performance

Figure 4 shows the current density – voltage (J-V) characteristics under 1 sun (100 mW/cm²) AM 1.5G illumination of planar heterojunction devices of the following structure: ITO/MoO₃ (30 nm)/ ClInPc (5 nm seed)/ClInPc (15 nm)/C₆₀ (40 nm)/BCP (10 nm)/Ag (100 nm). The first 5 nm ClInPc serves as a seed layer to control morphology of the additional 15 nm ClInPc. Observed performance parameters are summarized in Table 1. Devices with an annealed seed layer give better overall performance than devices with an as-deposited seed layer. PCE is gradually increased from 1.68% to 1.97% upon annealing. The greatest improvement is observed as the annealing temperature is increased from 60°C to 80°C. Annealing at 100°C induces

identical device performance as annealing at 80°C. This increase in device performance is concomitant with the evolution of a triclinic phase within the ClInPc films. Figure 5 shows the dependencies of performance parameters on seed layer annealing conditions. Improvements in J_{sc} and FF are observed upon annealing up to 80°C, while V_{oc} decreases with annealing temperature. Overall, the highest PCE is obtained for seed layers annealed at 80°C.

Figure 4. The J-V characteristics of ClInPc/C₆₀ planar solar cells containing seed layer as deposited or annealed under 1 sun simulated AM 1.5G solar illumination.

Figure 5. Comparison of the ClInPc/C₆₀ planar solar cell performance dependencies on the ClInPc seed layer annealing conditions.

Table 1. Solar cell performance parameters extracted from J-V characteristics in Figures 4 and 7.

Planar	100°C	5.3	0.63	0.59	1.97
Planar-mixed	As-deposited	4.0	0.64	0.40	1.02
Planar-mixed	80°C	5.5	0.59	0.57	1.85
Planar	60°C	4.6	0.71	0.54	1.76
Planar	80°C	5.2	0.64	0.59	1.96

Figure 6a shows the EQE and IQE spectra of the planar structure device with the seed layer as-deposited or annealed at 80°C. Considering the comparison between the devices comprising the same materials, mismatch factor is not included. Enhancement of EQE at regions below 700 nm and between 800-900 nm contributes to the increase of J_{sc} from 4.4 mA/cm² to 5.2 mA/cm². The increase at 800-900 nm corresponds to the increased triclinic phase absorption. Thus, transforming the ClInPc film from a monoclinic to triclinic phase leads to stronger absorption in the NIR region and enhances the EQE and PCE of solar cells.

IQE is also enhanced across the entire ClInPc photoactive region, suggesting improved charge collection due to morphological factors. One factor is the lower series resistance R_s in the device containing triclinic phase. The R_s of the device with seed layer as-deposited or annealed at 80°C is 22.99 Ω cm² and 14.99 Ω cm², respectively, as shown in Table 2. The reduced R_s presumably induces FF improvement as shown in Table 1. The reduced R_s arises because the stronger self-exchange coupling interaction⁵⁰⁻⁵² between the adjacent molecules in the triclinic phase leads to more efficient charge transport. Another reason that may contribute to increased IQE is the more delocalized electronic states³⁴ introduced by triclinic phase, which essentially benefits exciton diffusion length and exciton dissociation efficiency. Photoluminescence (PL) based exciton diffusion length study is not included in this work due to the unavailability of ClInPc PL signal. It is because ClInPc has the favorable combination of high dipole moments associated respectively with paramagnetism and the presence of a central heavy metal, which leads to an intersystem crossing in 300 ps with a quantum yield close to unity.⁵³ Hence, we studied the exciton diffusion length in the ClInPc/C₆₀ planar solar cells with seed layer with or without annealing. The results are shown in Figure 6b. The solid square dots are the experimental IQE values of the devices with the ClInPc thicknesses of 2, 5, 7, 15 and 20 nm. The J-V characteristics of the corresponding devices are shown in Figure S1, and their EQE and 1-Reflectance values are shown in Figure S2. The solid lines are the fitting curves following 1D diffusion equation:

$$IQE = (L_d/t) \times (1 - \exp(-2t/L_d)) / (1 + \exp(-2t/L_d)) \quad (1)$$

Where L_d is the exciton diffusion length, t is ClInPc film thickness and IQE is thickness dependent internal quantum efficiency.^{48,49} To fit the experimental IQE- t data, a value of $L_d = 2.4$ nm is applied for the device with an as-deposited seed layer, and a value of $L_d = 2.9$ nm is applied for the device with a seed layer annealed at 80°C. Therefore, a slightly increased exciton diffusing length is observed in the triclinic phase enriched device.

Figure 6. (a) The EQE and IQE spectra of the planar structure devices containing seed layers as-deposited or annealed at 80°C. (b) The thickness dependent experimental IQE values (solid square dots) and fitted 1D diffusion equation curve (equation 1).

An increased concentration of the triclinic phase in ClInPc films is accompanied by a photovoltage loss in devices. Several possible reasons can cause V_{oc} loss. Lower ΔE_{DA} (energy difference between the highest occupied molecular orbital (HOMO) of the donor and the lowest unoccupied molecular orbital (LUMO) of the acceptor) at triclinic phase ClInPc/ C_{60} interface compared to the monoclinic phase ClInPc/ C_{60} interface could possibly contribute to V_{oc} loss, since the HOMO of the triclinic phase is shallower than that of the monoclinic phase⁵⁴. Other possible reasons include the roughness introduced by the triclinic phase as discussed above and studied elsewhere, or heating-induced nonuniformity in organic films.

Here we examine the effect of the triclinic phase on the dark saturation current pre-exponential factor, which essentially determines V_{oc} ⁵⁰⁻⁵¹. Dependence of V_{oc} on the dark saturation current is revealed by equation 2:

$$V_{oc} \approx nkTq \ln J_{sc} J_s \quad (2)$$

where n is diode ideality factor, k is the Boltzmann constant, and q is elementary charge. The dark saturation current (J_s) is dominated by recombination ($n \approx 2$) and can be represented by equation 3:

$$J_s = J_{so} \exp(-\Delta E_{DA} / 2nkT) \quad (3)$$

Combining equations (2) and (3) yields

$$V_{oc} = (nkT/q) \ln(J_{sc}/J_s) + \Delta E_{DA} / 2q \quad (4)$$

The pre-exponential factor is a measure of ratio of charge generation to recombination. Extracted n , J_s and J_{so} values for devices containing seed layers as-deposited or annealed at 80°C are listed in Table 2. ΔE_{DA} is 1.3 eV for ClInPc/C₆₀ heterojunction based on HOMO of ClInPc 5.3 eV²⁰ and LUMO of C₆₀ 4.0 eV⁵⁵. J_{so} is 0.19 mA/cm² for the device with as-deposited seed layer and 0.49 mA/cm² for the device with seed-layer annealed at 80°C. The magnitude of J_{so} is often affected by the intermolecular reorganization energy for electron transfer from the donor to the acceptor, the electrical conductivity of organic films, the density of relevant states at D/A interface and the strength of coupling between the donor and acceptor material. Since stronger π interaction between adjacent ClInPc molecules in the triclinic phase compared to the monoclinic phase has been studied, it is possible that the same enhanced interaction at ClInPc and C₆₀ interface exists and facilitates bimolecular recombination, which leads to increased J_{so} . The calculated V_{oc} in Table 2 is derived by solving Eq. (4), where we assume ΔE_{DA} is the same for both devices. As a result, the difference between the calculated V_{oc} is identical to that of the measured V_{oc} , suggesting V_{oc} loss upon annealing is derived from the first item of Eq. (4), in other words, from J_{so} . Therefore, the tradeoff between enhanced charge generation and collection and simultaneously increased recombination derived from stronger intermolecular or interfacial interaction of triclinic phase should be carefully considered when engineering non-planar phthalocyanine solar cells.

To investigate if an annealed seed layer improves the performance of bulk heterojunction solar cells, planar-mixed solar cells were fabricated. The structure is ITO/MoO₃ (10 nm)/ClInPc (10 nm)/ClInPc:C₆₀ (1:1) (10 nm)/C₆₀ (20 nm)/BCP (10 nm)/Ag (100 nm), in which the first 5 nm of ClInPc is seed layer, either as-deposited or annealed at 80°C. As discussed above, the presence of the triclinic phase was revealed by the absorption spectrum (Figure 3c). Figure 7 shows the J-V characteristics of the planar-mixed solar cell. The device performance parameters are included in Table 1. J_{sc} is increased from 4.0 to 5.5 mA/cm². For the device with as-deposited seed layer, all J_{sc} , V_{oc} and FF values are lower than planar structure, largely due to prevalent bimolecular recombination as the donor-acceptor interface is increased. With an annealed seed layer, triclinic phase in bulk heterojunction helps increase exciton delocalization leading to increased charge collection. As a result, J_{sc} is increased from 4.0 mA/cm² to 5.5 mA/cm², FF is increased from 0.40 to 0.57, V_{oc} also shows decrease from 0.64 V to 0.59 V, and PCE exhibits a 69% increase. Thus we suggest that for non-planar phthalocyanine NIR absorbing solar cells, utilizing a seed layer to control the morphology of entire photoactive layer not only is applicable for planar architecture solar cell, but also can be a useful strategy for planar-mixed solar cell.

Figure 7. The J-V characteristics of ClInPc/ClInPc:C₆₀/C₆₀ planar-mixed solar cells containing seed layer as deposited or annealed at 80°C under 1 sun simulated AM 1.5G solar illumination.

Table 2. The relevant numbers from J-V characteristics of ClInPc/C₆₀ planar solar cell with seed layer as-deposited or annealed at 80°C in Figure 4 and Table 1. Calculated V_{oc} is taken from solving Eq. (4).

Annealing Condition	R_s ($\Omega \text{ cm}^2$)	n	J_s (mA/cm^2)	J_{s0} (mA/cm^2)	Obs. V_{oc} (V)	Calc. V_{oc} (V)
As-deposited	22.99	2.28	3.45×10^{-3}	0.19	0.72	0.83
80°C	14.99	2.06	2.61×10^{-3}	0.49	0.64	0.77

4. Conclusions

The morphologies of ClInPc neat films and ClInPc:C₆₀ mixed films were effectively controlled using a pre-deposited ClInPc seed layer. The triclinic phase of ClInPc is enriched in films grown on annealed seed layers, as revealed by the evolution of a red-shifted absorption shoulder compared to the monoclinic phase absorption. The same ClInPc neat films and ClInPc:C₆₀ mixed films were incorporated into planar and planar-mixed solar cells. A correlation between triclinic phase content and increased J_{sc} , FF and PCE, as well as reduced V_{oc} was observed. Increased J_{sc} , FF, EQE and IQE is attributed to the enhanced absorption at NIR region and reduced series resistance in the triclinic phase-containing films. Decreased V_{oc} is correlated to the higher J_{s0} . Both reduced series resistance and increased J_{s0} are derived from stronger intermolecular interaction (at donor/donor or donor/acceptor interface) in the triclinic form compared to the monoclinic form.

5. Acknowledgments

The authors gratefully acknowledge financial support from the Air Force Office of Scientific Research (Grant number FA9550-14-1-0128).

References

1. Spanggaard, H.; Krebs, F. C., A brief history of the development of organic and polymeric photovoltaics. *Solar Energy Materials and Solar Cells* **2004**, *83* (2–3), 125-146.
2. Krebs, F. C.; Gevorgyan, S. A.; Gholamkhass, B.; Holdcroft, S.; Schlenker, C.; Thompson, M. E.; Thompson, B. C.; Olson, D.; Ginley, D. S.; Shaheen, S. E.; Alshareef, H. N.; Murphy, J. W.; Youngblood, W. J.; Heston, N. C.; Reynolds, J. R.; Jia, S.; Laird, D.; Tuladhar, S. M.; Dane, J. G. A.; Atienzar, P.; Nelson, J.; Kroon, J. M.; Wienk, M. M.; Janssen, R. A. J.; Tvingstedt, K.; Zhang, F.; Andersson, M.; Inganäs, O.; Lira-Cantu, M.; de Bettignies, R.; Guillerez, S.; Aernouts, T.; Cheyns, D.; Lutsen, L.; Zimmermann, B.; Würfel, U.; Niggemann, M.; Schleiermacher, H.-F.; Liska, P.; Grätzel, M.; Lianos, P.; Katz, E. A.; Lohwasser, W.; Jannon, B., A round robin study of flexible large-area roll-to-roll processed polymer solar cell modules. *Solar Energy Materials and Solar Cells* **2009**, *93* (11), 1968-1977.
3. Brabec, C. J.; Gowrisanker, S.; Halls, J. J.; Laird, D.; Jia, S.; Williams, S. P., Polymer-fullerene bulk-heterojunction solar cells. *Adv Mater* **2010**, *22* (34), 3839-56.
4. Steim, R.; Kogler, F. R.; Brabec, C. J., Interface materials for organic solar cells. *J Mater Chem* **2010**, *20* (13), 2499-2512.
5. Heremans, P.; Cheyns, D.; Rand, B. P., Strategies for Increasing the Efficiency of Heterojunction Organic Solar Cells: Material Selection and Device Architecture. *Accounts of Chemical Research* **2009**, *42* (11), 1740-1747.
6. Hains, A. W.; Liang, Z.; Woodhouse, M. A.; Gregg, B. A., Molecular Semiconductors in Organic Photovoltaic Cells. *Chemical reviews* **2010**, *110* (11), 6689-6735.
7. Thompson, B. C.; Fréchet, J. M. J., Polymer–Fullerene Composite Solar Cells. *Angewandte Chemie International Edition* **2008**, *47* (1), 58-77.
8. Brédas, J.-L.; Norton, J. E.; Cornil, J.; Coropceanu, V., Molecular Understanding of Organic Solar Cells: The Challenges. *Accounts of Chemical Research* **2009**, *42* (11), 1691-1699.
9. You, J.; Dou, L.; Yoshimura, K.; Kato, T.; Ohya, K.; Moriarty, T.; Emery, K.; Chen, C.-C.; Gao, J.; Li, G.; Yang, Y., A polymer tandem solar cell with 10.6% power conversion efficiency. *Nature communications* **2013**, *4*, 1446.
10. Ameri, T.; Li, N.; Brabec, C. J., Highly efficient organic tandem solar cells: a follow up review. *Energy & Environmental Science* **2013**, *6* (8), 2390-2413.
11. Che, X.; Xiao, X.; Zimmerman, J. D.; Fan, D.; Forrest, S. R., High-Efficiency, Vacuum-Deposited, Small-Molecule Organic Tandem and Triple-Junction Photovoltaic Cells. *Advanced Energy Materials* **2014**, *4* (18), n/a-n/a.
12. Riede, M.; Uhrich, C.; Widmer, J.; Timmreck, R.; Wynands, D.; Schwartz, G.; Gnehr, W.-M.; Hildebrandt, D.; Weiss, A.; Hwang, J.; Sundarraj, S.; Erk, P.; Pfeiffer, M.; Leo, K., Efficient Organic Tandem Solar Cells based on Small Molecules. *Advanced Functional Materials* **2011**, *21* (16), 3019-3028.
13. Perez, M. D.; Borek, C.; Djurovich, P. I.; Mayo, E. I.; Lunt, R. R.; Forrest, S. R.; Thompson, M. E., Organic Photovoltaics Using Tetraphenylbenzoporphyrin Complexes as Donor Layers. *Advanced Materials* **2009**, *21* (14-15), 1517-1520.
14. Rand, B. P.; Xue, J.; Yang, F.; Forrest, S. R., Organic solar cells with sensitivity extending into the near infrared. *Applied Physics Letters* **2005**, *87* (23), 233508.
15. Bailey-Salzman, R. F.; Rand, B. P.; Forrest, S. R., Near-infrared sensitive small molecule organic photovoltaic cells based on chloroaluminum phthalocyanine. *Applied Physics Letters* **2007**, *91* (1), 013508.
16. Chauhan, K. V.; Sullivan, P.; Yang, J. L.; Jones, T. S., Efficient Organic Photovoltaic Cells through Structural Modification of Chloroaluminum Phthalocyanine/Fullerene Heterojunctions. *The Journal of Physical Chemistry C* **2010**, *114* (7), 3304-3308.

17. Huang, P.; Du J.; Gunathilake, S. S.; Rainbolt, E. A.; Murphy, J. W.; Black, K. T.; Barrera, D.; Hsu, J. P.; Gnade, B. E.; Stefan M. C.; Biewer, M. C., Benzodifuran and benzodithiophene donor–acceptor polymers for bulk heterojunction solar cells. *Journal of Materials Chemistry A* **2015**, *3*, 6980-6989.
18. Shea, M. J.; Arnold, M. S., 1% solar cells derived from ultrathin carbon nanotube photoabsorbing films. *Applied Physics Letters* **2013**, *102* (24), 243101.
19. Wu, M.-Y.; Jacobberger, R. M.; Arnold, M. S., Design length scales for carbon nanotube photoabsorber based photovoltaic materials and devices. *Journal of Applied Physics* **2013**, *113* (20), 204504.
20. Armstrong², W. W. a. N., Doping Effect on Chloroindium Phthalocyanine (ClInPc)/C60 Solar Cells. *Mater. Res. Soc. Symp. Proc.* **2012**.
21. Fleetham, T. B.; Mudrick, J. P.; Cao, W.; Klimes, K.; Xue, J.; Li, J., Efficient zinc phthalocyanine/C60 heterojunction photovoltaic devices employing tetracene anode interfacial layers. *ACS Appl Mater Interfaces* **2014**, *6* (10), 7254-9.
22. Hiramoto, M.; Kitada, K.; Iketaki, K.; Kaji, T., Near infrared light driven organic p-i-n solar cells incorporating phthalocyanine J-aggregate. *Applied Physics Letters* **2011**, *98* (2), 023302.
23. Kim, T.-M.; Kim, H. J.; Shim, H.-S.; Choi, M.-S.; Kim, J. W.; Kim, J.-J., The epitaxial growth of lead phthalocyanine on copper halogen compounds as the origin of templating effects. *Journal of Materials Chemistry A* **2014**, *2* (23), 8730.
24. Claessens, C. G.; Hahn, U.; Torres, T., Phthalocyanines: from outstanding electronic properties to emerging applications. *Chemical record* **2008**, *8* (2), 75-97.
25. Heutz, S.; Bayliss, S. M.; Middleton, R. L.; Rumbles, G.; Jones, T. S., Polymorphism in Phthalocyanine Thin Films: Mechanism of the $\alpha \rightarrow \beta$ Transition. *The Journal of Physical Chemistry B* **2000**, *104* (30), 7124-7129.
26. Achar, B. N.; Lokesh, K. S., Studies on polymorphic modifications of copper phthalocyanine. *Journal of Solid State Chemistry* **2004**, *177* (6), 1987-1993.
27. Ren, J.; Meng, S.; Kaxiras, E., Theoretical investigation of the C60/copper phthalocyanine organic photovoltaic heterojunction. *Nano Research* **2012**, *5* (4), 248-257.
28. Wim Dexters, E. B., Milos Nesladek, Jan D'Haen, Etienne Goovaertsc and Ken Haenen, Molecular orientation of lead phthalocyanine on (100) oriented single crystal diamond surfaces. *Phys.Chem.Chem.Phys.* **2015**, *17*, 9619.
29. Frank Herbstein, R. M., Changes in Space and Laue Groups of Some Published Crystal Structures. *Acta Cryst.* **1982**, *B38*, 1051.
30. J. Mizuguchi, G. R., H. R. Karfunkel, Solid-state Spectra of Titanylphthalocyanine As Viewed from Molecular Distortion. *J. Phys. Chem. A* **1995**, *99*, 16217.
31. Ukei, K., Lead phthalocyanine. *Acta Crystallographica Section B* **1973**, *29* (10), 2290-2292.
32. Iyechika, Y.; Yakushi, K.; Ikemoto, I.; Kuroda, H., Structure of lead phthalocyanine (triclinic form). *Acta Crystallographica Section B* **1982**, *38* (3), 766-770.
33. Kim, T. M.; Shim, H. S.; Choi, M. S.; Kim, H. J.; Kim, J. J., Multilayer epitaxial growth of lead phthalocyanine and C(70) using CuBr as a templating layer for enhancing the efficiency of organic photovoltaic cells. *ACS Appl Mater Interfaces* **2014**, *6* (6), 4286-91.
34. Varughese, S., Non-covalent routes to tune the optical properties of molecular materials. *Journal of Materials Chemistry C* **2014**, *2* (18), 3499.
35. Vasseur, K.; Broch, K.; Ayzner, A. L.; Rand, B. P.; Cheyns, D.; Frank, C.; Schreiber, F.; Toney, M. F.; Froyen, L.; Heremans, P., Controlling the texture and crystallinity of evaporated lead phthalocyanine thin films for near-infrared sensitive solar cells. *ACS Appl Mater Interfaces* **2013**, *5* (17), 8505-15.
36. Zawadzka, A.; Płóciennik, P.; Czarnecka, I.; Sztupecka, J.; Łukasiak, Z., The effects of annealing process influence on optical properties and the molecular orientation of selected organometallic compounds thin films. *Optical Materials* **2012**, *34* (10), 1686-1691.

37. K. V. Chauhan, P. S., J. L. Yang, T. S. Jones, Efficient Organic Photovoltaic Cells through Structural Modification of Chloroaluminum Phthalocyanine/Fullerene Heterojunctions. *J. Phys. Chem. C* **2010**, *114*, 3304.
38. Duan, H.; Yang, J.; Fu, L.; Xiong, J.; Yang, B.; Ouyang, J.; Zhou, C.; Huang, H.; Gao, Y., Interface modification of organic photovoltaics by combining molybdenum oxide (MoO_x) and molecular template layer. *Thin Solid Films* **2015**, *574*, 146-151.
39. Soo Yook, K.; Doo Chin, B.; Yeob Lee, J.; Lassiter, B. E.; Forrest, S. R., Vertical orientation of copper phthalocyanine in organic solar cells using a small molecular weight organic templating layer. *Applied Physics Letters* **2011**, *99* (4), 043308.
40. Shim, H.-S.; Kim, H. J.; Kim, J. W.; Kim, S.-Y.; Jeong, W.-I.; Kim, T.-M.; Kim, J.-J., Enhancement of near-infrared absorption with high fill factor in lead phthalocyanine-based organic solar cells. *J Mater Chem* **2012**, *22* (18), 9077.
41. Ngongang Ndjawa, G. O.; Graham, K. R.; Li, R.; Conron, S. M.; Erwin, P.; Chou, K. W.; Burkhard, G. F.; Zhao, K.; Hoke, E. T.; Thompson, M. E.; McGehee, M. D.; Amassian, A., Impact of Molecular Orientation and Spontaneous Interfacial Mixing on the Performance of Organic Solar Cells. *Chemistry of Materials* **2015**, *27* (16), 5597-5604.
42. Lim, X., The slow-chemistry movement. *Nature* **2015**, *524*.
43. Miyamoto, A.; Nichogi, K.; Taomoto, A.; Nambu, T.; Murakami, M., Structural Control of Evaporated Lead-Phthalocyanine films, *Thin Solid Films*, **1995**, *256*, 64-67.
44. Basova, T. V.; Kiselev, V. G.; Dubkov, I. S.; Latteyer, F.; Gromilov, S. A.; Peisert, H.; Chassè, T., Optical Spectroscopy and XRD Study of Molecular Orientation, Polymorphism, and Phase Transitions in Fluorinated Vanadyl Phthalocyanine Thin Films. *The Journal of Physical Chemistry C* **2013**, *117* (14), 7097-7106.
45. Vasseur, K.; Rand, B. P.; Cheyns, D.; Temst, K.; Froyen, L.; Heremans, P., Correlating the Polymorphism of Titanyl Phthalocyanine Thin Films with Solar Cell Performance. *The Journal of Physical Chemistry Letters* **2012**, *3* (17), 2395-2400.
46. Placencia, D.; Wang, W.; Gantz, J.; Jenkins, J. L.; Armstrong, N. R., Highly Photoactive Titanyl Phthalocyanine Polymorphs as Textured Donor Layers in Organic Solar Cells. *The Journal of Physical Chemistry C* **2011**, *115* (38), 18873-18884.
47. Placencia, D.; Wang, W.; Shallcross, R. C.; Nebesny, K. W.; Brumbach, M.; Armstrong, N. R., Organic Photovoltaic Cells Based On Solvent-Annealed, Textured Titanyl Phthalocyanine/C60Heterojunctions. *Advanced Functional Materials* **2009**, *19* (12), 1913-1921.
48. Ortí, E.; Brédas, J. L.; Clarisse, C., Electronic structure of phthalocyanines: Theoretical investigation of the optical properties of phthalocyanine monomers, dimers, and crystals. *The Journal of chemical physics* **1990**, *92* (2), 1228.
49. Telfer, S. G.; McLean, T. M.; Waterland, M. R., Exciton coupling in coordination compounds. *Dalton transactions* **2011**, *40* (13), 3097-108.
50. Erwin, P.; Thompson, M. E., Elucidating the interplay between dark current coupling and open circuit voltage in organic photovoltaics. *Applied Physics Letters* **2011**, *98* (22), 223305.
51. Perez, M. D.; Borek, C.; Forrest, S. R.; Thompson, M. E., Molecular and Morphological Influences on the Open Circuit Voltages of Organic Photovoltaic Devices. *Journal of the American Chemical Society* **2009**, *131* (26), 9281-9286.
52. Schlenker, C. W.; Thompson, M. E., The molecular nature of photovoltage losses in organic solar cells. *Chemical Communications* **2011**, *47* (13), 3702-3716.
53. James S. Shirk, R. G. S. P., Steven R. Flom, Heino Heckmann, and Michael Hanack, Effect of Axial Substitution on the Optical Limiting Properties of Indium Phthalocyanines. *J. Phys. Chem. A* **104**.
54. Akira J. Ikushima, T. K., Shuji Yoshida, Akihiro Maeda Valence and conduction band edges of metal-phthalocyanines and carrier behavior. *Thin Solid Films* **1996**, *273*, 35.

55. Peng, Y.; Zhang, L.; Andrew, T. L., High open-circuit voltage, high fill factor single-junction organic solar cells. *Applied Physics Letters* **2014**, *105* (8), 083304.