Effects of Na and K co-doping on growth and scintillation properties of Eu:SrI₂ crystals

Tomoki Ito¹, Yuui Yokota², Shunsuke Kurosawa^{1,2}, Robert Kral³, Jan Pejchal^{2,3}, Yuji Ohashi¹,
Kei Kamada^{2,4}, Martin Nikl³, Akira Yoshikawa^{1,2,4},

¹ Institute for Materials Research, Tohoku University

2-1-1, Katahira, Aoba-ku, Sendai, Miyagi, 980-8577, JAPAN

² New Industry Creation Hatchery Center (NICHe), Tohoku University

6-6-10, Aramaki, Aoba, Aoba-ku, Sendai, Miyagi, 980-8579, JAPAN

³ Institute of Physics AS CR

10 Cukrovarnicka, Prague 6, 16200 Czech Republic

⁴ C&A corporation

6-6-40, Aramaki, Aoba, Aoba-ku, Sendai, Miyagi, 980-8579, JAPAN

Corresponding author : Tomoki Ito

E-mail : tomoki@imr.tohoku.ac.jp

Abstract

We grew Na and K co-doped Eu:SrI₂ [Na,Eu:SrI₂ and K,Eu:SrI₂] crystals by a modified micro-pulling-down method to reveal the co-doping effects on the crystal growth and scintillation properties. The non-codoped, Na0.5%, Na1.0%, K0.5% and K1.0%,Eu:SrI₂ crystals indicated high transparency while the milky parts were generated in the Na5.0% and K5.0%,Eu:SrI₂ crystals. The light yields of Na,Eu:SrI₂ and K,Eu:SrI₂ crystals under γ -ray irradiation were decreased by the Na and K co-doping. On the other hand, there was a small change within 940~1020 ns in the decay times by the Na and K co-doping. In the light yield proportionality under γ -ray irradiation, the non-proportionality in the low energy region was improved by Na and K co-doping.

Keywords

Eu:SrI₂, Scintillator, Single crystal, Alkali metal, Light yield, Non-proportionality

1. Introduction

Eu doped SrI_2 [Eu:SrI₂] scintillator crystal has high light yield (~80,000 ph/MeV) and high energy resolution (~3%) due to the small band-gap. Therefore, it is expected to be applied for next-generation radiation detector (Cherepy et al., 2008, Belli et al., 2012, Glodo et al., 2010, Cherepy et al., 2009, Hawrami et al., 2013, Gahane et al., 2009). The light yield proportionality of Eu:SrI₂ single crystal has been measured in the previous reports (Cherepy et al., 2008) and the Eu:SrI₂ single crystal indicated excellent light yield proportionality compared to the other scintillators as represented by Tl:NaI, Ce:LaBr₃ and Ce:Y₃Al₅O₁₂. In our previous report [Yokota et al., 2013, Nishimoto et al., 2014], the light yield proportionality of La co-doped Eu:SrI₂ crystals was measured and the non-proportionality became worse especially in the low energy region (< 200 keV) by the La co-doping. On the other hand, there is a previous report about the effects of alkali metal (Na⁺) or alkali earth metal (Mg²⁺, Ca²⁺, Sr²⁺ and Ba²⁺) doping on the non-proportionality of the CeBr₃ scintillator crystal and the non-proportionality in the low energy region was improved by the doping (Quarati et al., 2014). According to the previous report, the improvement of the non-proportionality by the alkali metal or alkali earth metal doping is originated from a more efficient e-h pair creation and/or energy transfer from the e-h pairs to the luminescent centers at energy below 100 keV. However, there is no report about the effects of alkali metal (Na⁺ and K⁺) co-doping to Eu:SrI₂ crystals on the optical and scintillation properties. According to the result of CeBr₃ crystal, the improvement of the non-proportionality is also expected for the Eu:SrI₂ crystal by the alkali metal.

On these backgrounds, in this study, we grew Na and K co-doping Eu:SrI₂ crystals with various Na and K concentrations and measure the optical and scintillation properties to reveal the effects of the alkali metal co-doping on the optical and scintillation properties and improve the non-proportionality of Eu:SrI₂.

2. Experimental procedure

We grew the $\{(\text{Sr}_{1-x}\text{A}_x)_{0.925}\text{Eu}_{0.075}\}\text{I}_2$ [A,Eu:SrI₂, A = Na, K] crystals by a modified micro-pulling-down [μ -PD] method (Totsuka et al., 2012, Kral et al. 2013). Detail of the crystal growth procedure was described in the previous reports (Yokota et al. 2011). Mixed powders of the $\{(\text{Sr}_{1-x}\text{A}_x)_{0.925}\text{Eu}_{0.075}\}\text{I}_2$ with $x = 0.00, 0.005, 0.01$, and 0.05 were prepared from the starting materials of SrI₂ (4N), EuI₂ (5N), NaI (4N), and KI (4N) in the glove box filled with Ar (oxygen and moisture concentrations < 1 ppm). By the modified μ -PD method, we grew the Non-doped, Na,Eu:SrI₂ and K,Eu:SrI₂ fiber crystals in high-purity Ar gas (99.9999%) at

~0.05 mm/min growth rate using a Pt-Rh wire as a seed without rotation. Polished specimens with 1 mm thickness were prepared from the first half parts of the grown fiber crystals by cutting and polishing in a mineral oil for measurements of optical and scintillation properties.

Parts of the grown crystals were ground in the globe box for X-ray diffraction [XRD] measurement by using an X-ray diffractometer (BRUKER, D8 DISCOVER) and a sealed sample chamber with no diffraction peak in the 2θ range of 20-70°. The configuration of the sealed sample chamber was a 51.5 mm outer diameter half-sphere made of polystyrol and polymethyl methacrylate. The X-ray source was Cu- K_α with accelerating voltage of 40 kV and beam current of 40 mA). The Inductively Coupled Plasma [ICP] analysis (Thermo Fisher Scientific, IRIS Advantage DIO) of the as-grown crystals was performed to identify the actual chemical compositions.

Room temperature emission and excitation spectra of the polished specimens were measured by using a spectrophotometers A10094 and Hamamatsu KK (Kurosawa et al. 2014). Pulse-height spectra and the decay curves of the specimens under γ -ray excitation from ^{137}Cs radiation source were measured to estimate their light yields and decay times using a photomultiplier [PMT](HAMAMATSU, R7600U-200), multi-channel analyzer (AMPTEK CO. Pocket MCA 8000A) and an oscilloscope (TEKTRONIX, TBS1102). The measurements were performed in the grove box filled with Ar atmosphere, where the specimens and PMT were set. The shaping time was 2 μs in the pulse-height spectra measurements. The Ce 1 mol% doped Gd_2SiO_5 [Ce:GSO] single crystal with the light yield of 11,000 ph/MeV was used as a reference to estimate the light yields of the specimens. To evaluate the non-proportionality of the specimens, pulse-height spectra were measured with various γ -ray excitation sources (^{22}Na , ^{54}Mn , ^{60}Co , ^{133}Ba , ^{137}Cs , and ^{152}Eu). The non-proportionality of the Tl:NaI was also measured for the reference.

3. Results and discussions

Non-codoped $\text{Eu}:\text{SrI}_2$ and Na, $\text{Eu}:\text{SrI}_2$ and K, $\text{Eu}:\text{SrI}_2$ crystals with various Na and K concentrations were grown by the modified μ -PD method. It is shown in Fig.1(a) that the as-grown and polished crystals. All as-grown crystals were approximately $\phi 2$ mm in diameter and more than 15 mm in length. No visible cracks and inclusions were observed in the polished non-codoped $\text{Eu}:\text{SrI}_2$, Na0.5% and 1.0%, $\text{Eu}:\text{SrI}_2$, and K0.5% and 1.0%, $\text{Eu}:\text{SrI}_2$ crystals and these crystals showing high transparency. On the other hand, milky parts in the polished Na5.0%, $\text{Eu}:\text{SrI}_2$ and K5.0%, $\text{Eu}:\text{SrI}_2$ crystals were present. It is shown in Fig.1(b) that the powder XRD patterns of the as-grown crystals were measured for the phase identification. All diffraction peaks in the XRD patterns of the all crystals except for K5.0%, $\text{Eu}:\text{SrI}_2$ and Na5.0%, $\text{Eu}:\text{SrI}_2$ were identified by the SrI_2 crystal structure (Orthorhombic, Pnma). On the

other hand, the additional diffraction peaks of secondary phases were observed in the XRD pattern of the K5.0%,Eu:SrI₂ and Na5.0%,Eu:SrI₂ crystals. The secondary phases of K5.0%,Eu:SrI₂ crystal were considered to be KSr₂I₅ and KI while the secondary phase(s) in the Na5.0%,Eu:SrI₂ crystal was not identified. According to the ICP analysis, the actual Na and K concentrations of Na5.0%, K1.0% and K5.0%,Eu:SrI₂ crystals were 1.3%, 0.3% and 2.7%, respectively. The actual Na and K concentrations of Na0.5%, Na1.0% and K0.5%,Eu:SrI₂ crystals were the below the limit of detection as it is shown in Table 1. In the Na,Eu:SrI₂ and K,Eu:SrI₂ crystals, the actual Na and K concentrations were lower than the nominal composition due to the difference of the valences and ionic radius between Sr²⁺ and, Na⁺ and K⁺ ions. Therefore, another co-doping of the trivalence cation as represented by Y and La could be effective to improve the segregation of Na⁺ and K⁺.

We measured the room temperature photoluminescence spectra of the polished specimens (Fig.2). In the emission spectra performed 410 nm excitation (Fig.2(a)), an emission peak around 435 nm originated from 5d-4f transitions of Eu²⁺ ion was observed for all crystals except for K5.0%,Eu:SrI₂. The emission peak for K5.0%,Eu:SrI₂ crystal was shifted to higher wavelength by the K co-doping. However, the reason is not clear now. In the excitation spectra monitored at 435 nm, the excitation peak was observed around 422 nm for all crystals. In addition, a broad excitation peak appeared around 315 nm for K5.0%,Eu:SrI₂ crystal. Probably originating from the secondary phase of Eu:KI (Aceves et al. 2005).

It is shown in Fig.3 that the pulse-height spectra and decay curves of the non-codoped Eu:SrI₂, Na,Eu:SrI₂ and K,Eu:SrI₂ specimens under γ -ray irradiation. In the pulse-height spectra, clear photo-peaks were observed for the non-codoped, Na0.5%, Na1.0%, K0.5%, and K1.0%,Eu:SrI₂ specimens whereas there was no photo-peak in the Na5.0% and K5.0%,Eu:SrI₂ crystals. The light yields of the specimens were estimated by using the Ce:GSO crystal (11,000 ph/MeV) as a reference (Fig.3(a)). The estimated light yields of the non-codoped, K0.5%, K1.0%,Eu:SrI₂ specimens and Na0.5%, Na1.0%,Eu:SrI₂ specimens were 70,000, 46,000, 48,500, 53,500 and 49,000 ph/MeV, respectively, and the light yield of the Eu:SrI₂ was decreasing by the Na and K co-doping. In addition, the estimated energy resolutions of the non-codoped, K0.5%, K1.0%,Eu:SrI₂ specimens and Na0.5%, Na1.0%,Eu:SrI₂ specimens were 11.3, 34.8, 11.6, 20.0 and 39.1%, respectively. The energy resolutions were relatively worse than previous reports about Eu:SrI₂. The results are considered to be due to the degradation of the surface for the polished specimens. The scintillation decay curves were fitted by the single exponential decay function (Fig.3(b)) and the decay times were estimated. The decay times of all crystals were ranging from 940 to 1000 ns and there were few effects on the decay time with the Na and K co-doping as it was observed for the light yield evaluation as it is shown in Table 1.

The light yield proportionalities of the non-codoped Eu:SrI₂ and Na1.0%,Eu:SrI₂ and K1.0%,Eu:SrI₂ specimens were measured using various radiation sources emitting γ -ray as it is depicted in Fig.4. Comparing to the result of non-codoped Eu:SrI₂, the non-proportionalities of Na1.0%,Eu:SrI₂ and K1.0%,Eu:SrI₂ improved in the low energy region (< 200 keV). Especially, the effect of the improvement on the non-proportionality in the low energy region by the K co-doping was larger than that of the Na co-doping. The results suggest that alkali metal co-doping made the e-h pair creation and/or energy transfer from the e-h pairs to the luminescent centers at energy below 100 keV more efficient and the effect is dependent on the ionic radius of co-dopant.

4. Conclusions

We grew the Na and K co-doped Eu:SrI₂ crystals by the modified μ -PD method and their optical and scintillation properties were measured. The grown non-codoped Eu:SrI₂, Na,Eu:SrI₂ and K,Eu:SrI₂ crystals except for the Na5.0% and K5.0%,Eu:SrI₂. Grown crystals indicated high transparency while the milky parts were generated in the Na5.0% and K5.0%,Eu:SrI₂ crystals as indicated by presence of secondary phases in the powder XRD patterns. In the photoluminescence spectra, emission and excitation peaks were observed around 435 nm and 422 nm respectively for all specimens except for K5.0%,Eu:SrI₂. On the other hand, K5.0%,Eu:SrI₂ showed the emission peak at the longer wavelength and the additional excitation peak around 330 nm which was considered to be attributable to the secondary phases. In the pulse-height spectra, the light yield of the Eu:SrI₂ was decreased by the Na and K co-doping whereas the decay times were almost constant (940~1020 ns). Meanwhile, the non-proportionality of the the Eu:SrI₂ was improved in the low energy region (< 200 keV) by the Na and K co-doping and the K1.0%,Eu:SrI₂ crystal showed the most excellent non-proportionality.

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Figure Captions

Figure 1 (a) Non-codoped, Na,Eu:SrI₂ and K,Eu:SrI₂ crystals grown by the μ -PD method.
(b) Powder XRD patterns of the non-codoped, Na,Eu:SrI₂ and K,Eu:SrI₂ crystals.

Figure 2 (a) Emission and (b) excitation spectra of the non-codoped, Na,Eu:SrI₂ and K,Eu:SrI₂ specimens.

Figure 3 (a) Pulse-height spectra and (b) decay curves of the non-codoped, Na,Eu:SrI₂ and K,Eu:SrI₂ specimens under γ -ray irradiation.

Figure 4 Light yield proportionality of the non-codoped, Na,Eu:SrI₂ and K,Eu:SrI₂ specimens under γ -ray irradiation.

Table 1 Summary of chemical composition, and optical and scintillation properties for the Na,Eu:SrI₂ and K,Eu:SrI₂ crystals.

Figure 1

Figure 2

Figure 3

Figure 4

Table

Co-dopant	Nominal concentration of co-dopant / mol%	Actual concentration of co-dopant / mol% (measured by ICP)	Emission wavelength / nm	Excitation wavelength / nm	Light yield / photons· MeV ⁻¹	Energy resolution /%	Decay time /ns
K	0	< 0.1	435	420	70,000	11.3	959
	0.5	< 0.1	435	422	46,000	34.8	947
	1.0	0.3	436	421	48,500	11.6	941
	5.0	2.7	435	310, 424	-	-	968
Na	0.5	< 0.1	435	421	53,500	20.0	968
	1.0	< 0.1	435	422	49,000	39.1	1020
	5.0	1.3	435	424	-	-	947