Aromaticity and degree of aromatic condensation of char

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19 ABSTRACT

20 The aromatic carbon structure is a defining property of chars and is often
21 expressed with the help of two concepts: (i) aromaticity and (ii) degree of
22 aromatic condensation. The varying extent of these two features is assumed to
23 largely determine the relatively high persistence of charred material in the
24 environment and is thus of interest for e.g. biochar characterization or carbon
25 cycle studies. Consequently, a variety of methods has been used to assess the
26 aromatic structure of chars, which has led to interesting insights but has
27 complicated the comparison of data acquired with different methods. We
28 therefore used a suite of seven methods (elemental analysis, MIR spectroscopy,
29 NEXAFS spectroscopy, ^{13}C NMR spectroscopy, BPCA analysis, lipid analysis
30 and helium pycnometry) and compared 13 measurements from them using a
31 diverse sample set of 38 laboratory chars. Our results demonstrate that most of
32 the measurements could be categorized either into those which assess
33 aromaticity or those which assess the degree of aromatic condensation. A
34 variety of measurements, including relatively inexpensive and simple ones,
35 reproducibly captured the two aromatic features in question, and data from
36 different methods could therefore be compared. Moreover, general patterns
37 between the two aromatic features and the pyrolysis conditions were revealed,
38 supporting reconstruction of the highest heat treatment temperature (HTT) of
39 char.

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42 KEYWORDS

43 Pyrogenic organic matter, char, aromaticity, aromatic condensation, pyrolysis,
44 stability, heat treatment temperature, biochar

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1. Introduction

Natural and anthropogenic chars have recently received much attention (Knicker, 2011; Glaser and Birk, 2012; Manyà, 2012). Their role as important environmental constituents is increasingly being recognized; chars persist in soils and sediments, which has important implications for the global C budget (Schmidt and Noack, 2000) and they can exert beneficial properties on soils, improving fertility (Biederman and Harpole, 2013) and immobilizing hazardous compounds (Beesley et al., 2011). Moreover, anthropogenic chars (biochars) involve additional economic advantages; prudent biochar production can provide green energy, providing an interesting alternative to management of organic wastes (Meyer et al., 2011).

With increasing interest in the use of charred material, there is a growing need to characterize and classify the material accurately in order to improve understanding of its properties and behavior in the environment. A defining property of chars and of pyrogenic organic matter in general (Preston and Schmidt, 2006) is their aromatic C structure (Lehmann and Joseph, 2009), which is believed to consist of at least two different aromatic C phases: (i) an amorphous phase comprising randomly organized aromatic rings and (ii) a crystalline phase, comprising condensed polyaromatic sheets that are turbostratically aligned (Franklin, 1951; Cohen-Ofri et al., 2006; Keiluweit et al., 2010). The concepts of aromaticity (the total proportion of aromatic C including both phases; McNaught and Wilkinson, 1997) and that of the degree

of aromatic condensation (the proportion of the condensed aromatic C only; McBeath et al., 2011) relate to this two phase model. The varying extent of the two phases is believed to largely determine stability of the charred material against degradation in the environment (Lehmann et al., 2009; Singh et al., 2012). Consequently, aromaticity and the degree of aromatic condensation of a char likely influence its sequestration potential as well as the duration during which it can provide benefits to the soil (Nguyen et al., 2010).

The two features are themselves influenced by the feedstock, and the pyrolysis conditions. The type of feedstock contributes to the aromaticity and the degree of aromatic condensation by providing different chemical structures as starting material. For example, a high amount of aromatic structures in a feedstock (e.g. lignin in wood) can promote the resulting char aromaticity (Antal and Grønli, 2003). Similarly, different precursor materials attain a high degree of aromatic condensation at different temperatures (Setton et al., 2002). The pyrolysis conditions, in particular the highest heat treatment temperature (HTT), but also residence time, O₂ availability and pressure influence the C properties of the resulting char (Shafizadeh, 1982; Lua et al., 2004). Aromaticity has been reported to increase with HTT from 200 °C to ca. 500 °C, where maximum aromaticity values are reached. The degree of aromatic condensation showed, on the other hand, a more gradual increase with HTT from 400 °C, reaching maximum values at > 1000 °C (McBeath et al., 2011; Schneider et al., 2011).

Considering the importance of the aromatic C structure in char and its dependency on many influencing factors, it is not surprising that various attempts have been made to measure these archetypal properties of char. A wide variety of chemical and physical methods has been used, including elemental analysis, molecular markers, solid state ^{13}C nuclear magnetic resonance (NMR) spectroscopy, infrared (IR) spectroscopy, Raman spectroscopy, pyrolysis-gas chromatography-mass spectrometry (Py-GC-MS), X-ray diffraction, near edge X-ray absorption fine-structure spectroscopy (NEXAFS), X-ray photoelectron spectroscopy, measurements of surface area, He based solid density, electrical resistivity or high resolution transmission electron microscopy (HRTEM) (Derenne et al., 2005; McBeath et al., 2011; Charon et al., 2014).

While the wealth of methods for aromatic structure characterization of char is fascinating, with the methods continuing to grow in number and quality, it has become increasingly difficult to compare data using different methods and to relate the findings from them. Matters are complicated by the fact that terms such as aromaticity can have multiple technique-specific meanings (McBeath et al., 2011). Moreover, from a practical point of view, researchers and practitioners may have limited resources and instrumentations and would like to optimize both insightful data acquisition and reasonable analysis cost. Efforts to compare different methods and their measurements and to put them in a common framework are therefore required, thereby guiding the

interpretation of differently acquired data and suggesting suitable methods for specific analysis problems.

Here, we have used an extensive suite of 7 different methods that provided 13 measurements (cf. Table 1) in a comparative study to evaluate their assessment of the aromatic structure in charred materials. A large sample set, consisting of 38 different laboratory char samples (cf. Table 2), was analyzed using each method. The objective was to show and discuss how the 7 methods and their measurements capture the aromatic structures in the largely different chars and how they compare with each other. Moreover, we aimed to reveal the influence of feedstock, HTT and other pyrolysis parameters on aromaticity and degree of aromatic condensation, by statistically exploring the rich dataset from this large scale comparative study.

2. Material and methods

2.1 Char thermosequences

Four different feedstocks (chestnut wood, pine wood, fescue grass and rice grass) and two different pyrolysis procedures (A, 5 h residence time at HTT with continuous N₂ flow; B, 1 h residence time at HTT in a closed chamber) were used to create 38 laboratory char samples with HTT between 100 and 1000 °C (A, 200 – 1000 °C; B, 100 – 700 °C), as reported in detail in two previous studies (Keiluweit et al., 2010; Schneider et al., 2011). The resulting char thermosequences (Wood-A, Grass-A, Wood-B, Grass-B) are displayed in Table 2.

2.2 Methods, measurements and derived indices

The char samples were measured using elemental analysis, mid-IR (MIR) spectroscopy, NEXAFS spectroscopy, ¹³C NMR spectroscopy, benzene polycarboxylic acid (BPCA) analysis, lipid analysis and He pycnometry, as described in detail in the Supplementary data. Some of the methods provided more than one measurement (e.g. elemental analysis gives H/C and O/C ratios), as summarized in Table 1. Because the measurements are in widely different units and sometimes inversely proportional to each other, indices were built for comparison purposes that project the data on a common scale (0 – 1) and in the same direction (0, lowest aromaticity or lowest degree of aromatic condensation; 1, highest aromaticity or highest degree of aromatic

condensation). Thus, each value x of a measurement M was transformed according to Eq. 1 to provide $M_{index\ [0-1]}$. In the case of inversely proportional measurements, denoted $[1-0]$, they were further transformed according to Eq. 2 to yield $M_{index\ [0-1]}$. From here on, the methods and measurements are referred to by their indices (M_{index}), as listed in Table 1.

$$M_{index\ [0-1]\ or\ [1-0]} = \frac{x_M - \min_M}{\max_M - \min_M} \quad (1)$$

$$M_{index\ [0-1]} = 1 - M_{index\ [1-0]} \quad (2)$$

2.3. Data presentation and statistical analysis

The large set of original data can be found in the Supplementary data, while we focus here on the most important findings, showing only simplified data for reasons of clarity and understanding. All data analysis was conducted using the statistical software R (2011). The few missing values (ca. 5%) were imputed with the missForest algorithm (Stekhoven and Buehlmann, 2012) with an estimated normalized root mean square error of only 0.2%. Principal component analysis (PCA; Mardia et al., 1979) including biplots (Gabriel, 1971) was computed on the correlation matrix of the inputted dataset.

3. Results and discussion

All the methods provide measurements of physical or chemical variables linked either to aromaticity or degree of aromatic condensation, but the link between the variables and the derived aromatic features varies greatly. Measuring ring current with NMR (NMR-cond_{index}) is, for example, a more direct approach for assessing the degree of aromatic condensation than measuring skeletal density of char with He pycnometry (Pycno_{index}). However, the methods differ not only in how they provide assessment of aromatic structures but also in cost, availability and necessary expertise. Moreover, sample amount and composition can prevent the use of one method and/or promote the application of another. For these reasons, the aim was not to identify a reference method via inferential statistics or cost benefit analysis but rather to show method performance and explore qualitatively method similarities and differences within the two categories of aromaticity and aromatic condensation. The standardized measurement results from each method (indices) are shown in full in Figs. 1 and 2 and mathematically summarized according to their resemblance in form of a PCA, in Fig. 3.

3.1. Aromaticity

Aromaticity in char is generally easier to assess than the degree of aromatic condensation (McBeath et al., 2011). Thus, more methods are available for measuring aromaticity than for measuring the degree of aromatic

condensation (Figs. 1 and 3). In particular, the determination of elemental composition ($O-C_{index}$, $H-C_{index}$), the spectroscopic assessment of functional groups (MIR_{index} , $NEXAFS-aroma_{index}$, $NMR-aroma_{index}$) and the measurement of aromatic molecular markers ($BPCA_{index}$, PAH_{index}) can yield insight into the proportion of aromatic C in chars (Table 1). Non-aromatic molecular markers (TLE_{index} , ACL_{index}) may also be related to the proportion of aromatic C in chars as charring also affects the composition of these moieties.

The $O-C_{index}$ and the $H-C_{index}$ are routine measurements for chars (Baldock and Smernik, 2002; Hammes et al., 2006) and are an indirect measurement of aromaticity because the proportion of C in a sample increases with HTT, as dehydration, depolymerization and volatilization take place, eventually leading to the formation of H and O depleted aromatic C structures (Keiluweit et al., 2010; Wang et al., 2013).

The $O-C_{index}$ of fresh chars depicts a characteristic trend for aromaticity measurements (Fig. 1), because it traces closely the median of all aromatic measurements and thereby shows a typical pattern for aromaticity in the samples: The proportion of aromatic rings in a sample is lowest for low temperature chars, then increases sharply between 200 and 500 °C and reaches a plateau > 600 °C. The increasing $O-C_{index}$ reflects the elimination of O, mostly early in the charring process. The $O-C_{index}$ also differentiates between wood and grass char in the low temperature range (e.g. Wood-B vs. Grass-B). This can be attributed to the higher lignin content of the wood feedstock

(Rutherford et al., 2012). At high temperatures, the wood char samples again acquire higher $O\text{-}C_{\text{index}}$ values than the grass char samples. This probably reflects the higher ash content of the latter, which tends to retard aromatization (Mukome et al., 2013). Despite the fact that O content was assessed differently between the different thermosequences (actual measured O content for the A thermosequences vs. calculated by difference for the B thermosequences – cf. Supplementary data), similar results were obtained in both cases, indicating that the less expensive option of calculating O content by difference may be sufficient for assessing aromaticity in most freshly produced chars. However, measuring O by difference could be problematic for aromaticity assessment of chars with high ash contents.

The $H\text{-}C_{\text{index}}$ showed the lowest values for aromaticity in the lowest HTT range at 100 – 200 °C, as expected (Wang et al., 2013). Like the $O\text{-}C_{\text{index}}$, it also yields higher aromaticity for the lignin-rich wood char samples (Wood-B) than the lignin-poor grass char samples (Grass-B) at these low temperatures. A steep increase in aromaticity then followed from 200 – 500 °C HTT. However, highest values for $H\text{-}C_{\text{index}}$ based aromaticity were only reached at the highest HTT (1000 °C), indicating that the $H\text{-}C_{\text{index}}$ is not a pure measurement for aromaticity. In fact, the $H\text{-}C_{\text{index}}$ is positioned at the edge of the aromaticity measurements in Fig. 3 (vector number 9), relatively adjacent to the aromatic condensation measurements. This also makes sense from a theoretical point of view, because H is also lost during condensation of aromatic structures at high

charring temperature (Franklin, 1951). Thus, the $H-C_{\text{index}}$ probably indicates both an increasing amount of aromatic rings, up to ca. 500 °C HTT, and then the increasingly condensed nature of the aromatic rings, that continues to develop at higher charring temperature.

There was a slight lag in aromaticity of the Wood-B and Grass-B thermosequences vs. the Wood-A and Grass-A thermosequences with respect to HTT; the two thermosequences that were charred for 5 h and with a N_2 flow (A) attain higher values of aromaticity at slightly lower HTT than the two thermosequences that were charred for only 1 h in closed chambers (B). This pattern was observed for most of the aromaticity measurements (cf. $H-C_{\text{index}}$, $NMR\text{-aroma}_{\text{index}}$, $BPCA_{\text{index}}$, TLE_{index}) and is independent of feedstock (grass vs. wood). Thus, it appears to be related to the pyrolysis method, indicating that the longer residence times of pyrolysis procedure A charred the samples more completely at the same HTT than the shorter residence times in procedure B. The slight difference in pyrolysis procedures proved useful in this study because it allowed us to check if the various methods detect this feature.

The MIR_{index} distinguished low temperature chars from high temperature chars, with an increase in aromaticity between 300 and 500 °C, as similarly observed by Harvey et al. (2012a). It can thus serve as a quick and relatively inexpensive screening method for categorizing unknown char samples roughly according to their aromaticity and HTT. However, the MIR_{index} used here only qualifies as a rough estimate for aromaticity, given its large variability for

chars > 300 °C, which can also be seen from the shorter length of vector 5 in Fig. 3. Moreover, the MIR_{index} seems to depend on the pyrolysis procedure in this mid- and high-temperature range, which was not reproduced by any of the other methods. The data show that MIR has the potential to measure aromaticity and possibly other HTT dependent variables, but clearly requires more research and fine tuning with respect to IR band selection and spectral data analysis.

The NEXAFS-aroma_{index} has been similarly used to describe the B thermosequence chars elsewhere (Keiluweit et al., 2010) and was in the frame of the comparative study now also applied to part of the A thermosequence chars. In the case of the A thermosequence, the NEXAFS-aroma_{index} increased between 600 and 1000 °C. This is unlike most other aromaticity measurements that show a pronounced increase in aromaticity at lower HTT. The NEXAFS-aroma_{index} also gave higher aromaticity values for the Grass-A chars than the Wood-A chars; this is in contrast to most other observations but in line with previous results from NEXAFS applications (Keiluweit et al., 2010). In the case of the B thermosequence, the NEXAFS-aroma_{index} behaved unexpectedly in the low temperature range, indicating high aromaticity values for almost unaltered feedstock and showing decreasing aromaticity with increasing HTT. The reason for the unexpected behavior of the NEXAFS indices is discussed below in Section 3.2.

Using NMR is a recently established means for assessing aromaticity of charred samples (McBeath et al., 2011). The NMR-aroma_{index} determined the wood char samples in the low-temperature range (100 – 200 °C) as being more aromatic than the grass char samples, like the H-C_{index} and O-C_{index}, and then also showed the typical increase in aromaticity between 200 and 500 °C, reaching a plateau from 600 °C for all feedstocks and pyrolysis methods, with little variability. Since NMR measures directly the organic C bonds, it is not sensitive to ash content, yielding equally high values for high temperature wood and grass chars. A small shift between the differently pyrolysed thermosequences (A vs. B) could also be observed where aromaticity increased, further, corroborating the idea that the two pyrolysis procedures produced slightly different chars in the mid-temperature range.

The BPCA method separates and quantifies aromatic moieties that originate from condensed aromatic structures, which is why it is insensitive to lignin. Thus, the BPCA-aroma_{index} scored very low for both wood and grass chars in the range up to 200 °C. It then increased steadily with HTT up to 500 °C, where maximum values were reached, and decreased slightly from 700 °C onwards. The trend at low and mid-temperatures was consistent with the other aromaticity measurements, while the slight decrease at high temperature was inconsistent with the expected high aromaticity at high HTT. This can be attributed to a method artifact from which BPCA-aroma_{index} suffers: highly condensed aromatic structures are not completely converted to quantifiable

BPCAs and thus, as condensation increases above 600 °C, the BPCA-aroma_{index} declines slightly (Schneider et al., 2010). Still, it is able to distinguish the high temperature chars (grass vs. wood) and also the effect of the two different pyrolysis procedures A and B, which is why the BPCA-aroma_{index} plots well within the aromaticity vector in Fig. 3.

The lipid indices (TLE_{index}, ACL_{index}, PAH_{index}) are indirectly linked to changes in aromaticity because the lipid proportion (e.g. TLE_{index}) and quality (e.g. ACL_{index}, PAH_{index}) vary with HTT and the associated changes in aromaticity. The TLE_{index} showed a typical pattern for the aromaticity measurements, with low values (high extractability) at low temperatures and high values (low extractability) > 600 °C. Simple lipid extraction thus seemed to be a good proxy for aromaticity. It was able to differentiate between the two pyrolysis methods, showing the delayed increase in aromaticity of the B thermosequences, and remained at a constant maximum > 600 °C. However, it suffered from too high index values at low HTT for all thermosequences except the Grass-A thermosequence. This is probably due to the generally low content of extractable lipids in woody tissue (Gocke et al., 2013).

The ACL of alkanes decreased with increasing HTT due to cracking of carbon bonds (Simoneit and Elias, 2000; Wiesenberger et al., 2009), which translates into an increasing trend in the reciprocal ACL_{index} with increasing temperature. The ACL_{index} showed the general pattern of the other aromaticity measurements, e.g. pronounced increase in the index between 200 and 600 °C,

a higher lignin-derived aromaticity for woody low temperature chars than grassy chars and the characteristic difference between the two pyrolysis procedures between 200 and 600 °C. However, while the ACL_{index} was similar to other aromaticity measurements for the Grass-A thermosequence, it showed a high variability between the different thermosequences and sometimes no characteristic trend was observable at all (e.g. for the Wood-A thermosequence). These important drawbacks do not seem to depend on pyrolysis method or two types of starting material (wood vs. grass) but rather seem dependent on each of the four individual feedstocks. Thus, using ACL_{index} as an aromaticity measurement requires a careful calibration according to feedstock before conclusions can be drawn from it.

The PAH_{index} is another measurement theoretically linked to aromaticity in chars and which can be retrieved after lipid extraction (Simoneit, 2002; Yunker et al., 2002). Despite its rather straightforward concept of comparing larger with smaller PAH structures, no monotonic trend with HTT was observed. Instead, a maximum of the PAH_{index} was found at 400 – 500 °C, which interestingly correlates with total solvent-extractable PAH yield, as found in a more detailed study by Keiluweit et al. (2012). These results indicate that the relationship between PAH composition and HTT of chars is rather complex (Wiedemeier et al., 2014) and requires more research.

3.2. Degree of aromatic condensation

We assessed the degree of aromatic condensation by investigating functional groups and atomic bonds (NMR-cond_{index}, NEXAFS-cond_{index}), quantifying molecular markers (BPCA-cond_{index}) and measuring the structural density (Pycno_{index}) of the chars (Table 1).

The NMR method is the most recent development in the field and had already afforded plausible results on one set of the chars (Wood-A thermosequence; McBeath et al., 2011). In the same study, it was shown that the degree of aromatic condensation increases more gradually with increasing HTT than aromaticity and that a high degree of condensation only occurs at high HTT. Here, the inclusion of the Grass-A, Wood-B and Grass-B thermosequences again showed a similar pattern: the degree of aromatic condensation was minimal up to 250 °C and then constantly and monotonically increased up to the highest HTT. Interestingly, the chars from the B pyrolysis procedure again lagged behind those from the A pyrolysis procedure, indicating that the shorter pyrolysis time not only reduced the aromaticity but also the degree of aromatic condensation.

The BPCA method reproduced the NMR-cond_{index} general pattern consistently, despite the large difference in methodology. The relationship between the degree of aromatic condensation and HTT was more curved than that for NMR, pointing to a slightly non-linear behavior of the BPCA-cond_{index}. This minor deviation is probably method inherent because it occurred

reproducibly for all the thermosequences. However, it is surprising that the grass chars scored higher with respect to aromatic condensation than the wood chars at high temperature. This probably reflects the lower BPCA yield from grass chars than wood chars at high temperature (cf. Fig. 1, BPCA-aroma_{index}). This effect therefore has to be taken into account when the BPCA-cond_{index} is used for the assessment of the degree of aromatic condensation in chars from different feedstocks.

He pycnometry represents a third, completely independent approach for assessing the degree of aromatic condensation. He can penetrate the smallest of pores and can thus be used to assess the skeletal density (Brewer et al., 2014), which increases with increasing molecular condensation. Despite its lower cost and shorter analysis time, this indirect measurement appeared to reproduce quite well the main trend from the above, more complex and expensive aromatic condensation measurements. It depicted a rather linear increase in aromatic condensation from 300 °C up to the highest measured HTT. He pycnometry measurements in the low temperature range (< 300 °C) seem, however, to be less promising because the Pycno_{index} overestimated aromatic condensation for such mild heat treatment. Besides, the Wood-B thermosequence showed unusual behavior, whereby only the highest HTT (700 °C) would be assigned as giving a condensed aromatic carbon structure.

The NEXAFS-cond_{index} indicated a pronounced increase in the degree of aromatic condensation of the chars between 200 and 500 °C and a decrease

above 600 °C, which is not consistent with the other aromatic condensation measurements (Fig. 3, vector 2). Both NEXAFS indices used here (NEXAFS- $\text{aroma}_{\text{index}}$ and NEXAFS- $\text{cond}_{\text{index}}$) showed different behavior compared with other metrics of aromaticity and aromatic condensation. The use of simple indices derived from NEXAFS spectra is most likely limited by the fact that resonances in NEXAFS are not derived exclusively from specific functional groups but rather are a result of the entire electronic structure of the char material.

3.3. General patterns of aromatic features and implications

Applying a large suite of 7 methods with 13 measurements to a diverse char sample set of four thermosequences enabled us to identify general patterns between analytical methods and char C properties.

One observable pattern was the distinction between measurements that assess aromaticity and those that assess the degree of aromatic condensation, as discussed above and shown in Fig. 3. The differentiation was sharp and resulted in two distinct trends with HTT as depicted in Fig. 1 and Fig. 2. The two general trends across methods is best represented by the median of all measurements for each category: aromaticity increased sharply from 200 °C on, reaching maximum values at 500 – 600 °C, and stayed constant at the maximum with higher HTT (Fig. 1, bottom right). The median of all aromatic condensation measurements, on the other hand, increased smoothly from 300

°C on, reaching highest values at 1000 °C (Fig. 2, bottom left). While this difference between the two categories of aromatic measures has been reported elsewhere (McBeath et al., 2011), it was somewhat surprising how congruently the methods behaved within each category, despite the fact that they are based on fundamentally different approaches. The O-C_{index}, for example, followed the NMR-aroma_{index}, with only very minor differences for all thermosequences, and the H-C_{index}, as well as the BPCA-aroma_{index}, gave very similar results. Likewise, NMR-cond_{index}, BPCA-cond_{index} and Pycno_{index} were very similar for the degree of aromatic condensation for all feedstocks and pyrolysis procedures. Even the slight shift between pyrolysis procedure A and B with respect to HTT was readily detected with most of the methods (O-C_{index}, H-C_{index}, NMR-aroma_{index}, BPCA-aroma_{index}, TLE_{index}, NMR-cond_{index}, BPCA-cond_{index}, Pycno_{index}) reflecting both the importance of the pyrolysis conditions on the char C properties and the high sensitivity of each measurement for differences in char aromatic structure. Our observations imply that a variety of measurements are suitable for assessing the aromatic C structure of chars and that differently acquired data, e.g. from different research groups and laboratories, can be compared if the data are reasonably standardized and calibrated. More comparable data interpretations of diverse published and future studies alike can lead to a more consistent and better understanding of the C structure in different chars, which, in turn, promises to illuminate the C

sequestration potential of diverse chars (Wang et al. 2013; Keiluweit et al., 2010; Harvey et al., 2012b; Singh et al., 2012).

The distinctive and reproducibly measurable trends in aromaticity and aromatic condensation as a function of HTT also allow the inverse examination of chars: if the aim is to determine the unknown HTT of a certain char, it can be estimated by measuring its aromatic C structure. A quick multiple linear regression analysis of our dataset (data not shown) indicated that it is best to combine a measurement of aromaticity with one of aromatic condensation to obtain precise HTT reconstruction estimates. Aromaticity and aromatic condensation increase over different ranges of temperature and thus yield complementary information about the char HTT. Methods that provide both aromatic measurements at the same time (e.g. NMR-aroma_{index} and NMR-cond_{index} or BPCA-aroma_{index} and BPCA-cond_{index}) are thus beneficial from this point of view. However, similarly good results for HTT reconstruction were achieved by combining different methods, e.g. the O-C_{index}, which captures aromaticity, with the Pycno_{index}, which measures the degree of aromatic condensation. HTT reconstruction of chars may be an important application of the above methods in archeological studies (Conedera et al., 2009; Wolf et al., 2013) in addition to the more established focus on char stability in an environmental context (Nguyen et al., 2010; Al-Wabel et al., 2013).

We conclude that some indirect, relatively inexpensive and simple methods (e.g. elemental analysis or He pycnometry) captured the aromatic structures in

453 fresh chars similarly as well as more costly and time-consuming techniques.
454 This holds promise, whereby aromaticity and the degree of aromatic
455 condensation can be readily assessed in all kinds of char studies and with a
456 variety of new and established methods. Thus, characterization and
457 classification of charred materials should explicitly take these two C aromatic
458 features into account because they are likely key to their stability against
459 degradation in the environment as well as being .informative about production
460 temperature.

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468 Figure 1.

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470 **Figure 2.**

Figure 3.

FIGURE CAPTIONS

Fig. 1. Char aromaticity as measured via nine different indices. The median of all the indices is shown at the lower right corner and thus describes the typical trend in aromaticity with HTT.

Fig. 2. Degree of aromatic condensation as measured via four different indices. The median of all the aromatic condensation indices is depicted in the lower left and thus shows a typical increase in degree of aromatic condensation with HTT.

Fig. 3. PCA of the large dataset generated. The first PC is projected along the HTT axis and thus differentiates the chars according to heating temperature. The darker the circle symbol for a sample, the higher its HTT (top). The second PC spreads between aromaticity and the degree of aromatic condensation, showing that the char samples first become more aromatic with increasing HTT and then increase in aromatic condensation in the higher temperatures range (top). Likewise, the PCA differentiates between the different measurements (center and bottom) and groups them into typical aromaticity indices (3 – 9) and typical indices for assessing the degree of aromatic condensation (10 – 12). The numbers indicate: 1, PAH_{index}; 2, NEXAFS-

495 cond_{index}; 3, BPCA-aroma_{index}; 4, TLE_{index}; 5, MIR_{index}; 6, O-C_{index}; 7, NMR-
496 aroma_{index}; 8, ACL_{index}; 9, H-C_{index}; 10, BPCA-cond_{index}; 11, NMR-cond_{index}; 12,
497 Pycno_{index}; 13, NEXAFS-aroma_{index}.

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499 TABLES

500 **Table 1**

501 Methods considered comparatively for assessing aromaticity and/or degree of aromatic condensation of chars
 502 (names of derived indices in bold).

Measurement principle	Method	Index (M_{index})	Measurement	Reference
Elemental composition	Elemental analysis	$\text{O-C}_{\text{index}}$ $\text{H-C}_{\text{index}}$	C, H and O content (%): H/C, O/C	Baldock and Smernik (2002); Hammes et al. (2006)
Functional groups	Mid-infrared spectroscopy (MIR)	$\text{MIR}_{\text{index}}$	Aromaticity ratio (%): $(1420 + 821) / (1510 + 1320) \text{ cm}^{-1}$	Wood (1988); Guo and Bustin (1998); Moore and Owen (2001)
	Near-edge X-ray absorption fine structure spectroscopy (NEXAFS)	$\text{NEXAFS-aroma}_{\text{index}}$ $\text{NEXAFS-cond}_{\text{index}}$	Aromaticity ratio (%): 285 eV/ 286-288 eV Degree of condensation ratio (%): 284 eV/ 285eV	Francis and Hitchcock (1992); Agren et al. (1995); Kuznetsova et al. (2001); Brandes et al. (2008); Keiluweit et al. (2010)
	^{13}C Nuclear magnetic resonance spectroscopy (NMR)	$\text{NMR-aroma}_{\text{index}}$	Deconvolution on fitted spectrum with assigned aromatic peaks	McBeath and Smernik (2009); McBeath et al. (2011)
	- with sorbed ^{13}C label	$\text{NMR-cond}_{\text{index}}$	$-\Delta\delta$ (ppm): Shift of sorbed ^{13}C labelled benzene - shift of benzene	
Molecular markers	Benzene polycarboxylic acid (BPCA) analysis	$\text{BPCA-aroma}_{\text{index}}$ $\text{BPCA-cond}_{\text{index}}$	Total BPCA amount per organic carbon (g kg^{-1}): BPCA/C Ratio of B6CA per total BPCA amount (%): B6CA/BPCA	Schneider et al. (2011); Wiedemeier et al. (2013)
	Lipid analysis	$\text{TLE}_{\text{index}}$	Total lipid extract yield (g): TLE	Wiesenberg et al. (2009); Wiesenberg et al. (2010); Wiedemeier et al. (2014)
	- <i>n</i> -alkanes	$\text{ACL}_{\text{index}}$	Average chain length: ACL	
	- polycyclic aromatic hydrocarbons	$\text{PAH}_{\text{index}}$	Ratio of 4-6 ring to 2-3 ring polycyclic aromatic hydrocarbons (%)	
Density	He pycnometry	$\text{Pycno}_{\text{index}}$	Skeletal density (g cm^{-3})	Brown et al. (2006); Brewer et al. (2009); Brewer et al. (2014)

503

Table 2

Thermosequences measured with each method.

Feedstock	Pyrolysis Procedure	HTT	n	Thermosequence
Chestnut (Wood) (<i>Castanea sativa</i>)	A: 5 h HTT, N ₂ flow	200 - 1000 °C	12	Wood-A
Rice (Grass) (<i>Oryza sativa</i>)		200 - 1000 °C	12	Grass-A
Pine (Wood) (<i>Pinus ponderosa</i>)	B: 1 h HTT, Closed Chamber	100 - 700 °C	7	Wood-B
Fescue (Grass) (<i>Festuca arundinacea</i>)		100 - 700 °C	7	Grass-B

511

512 **Author Contributions**

513 D.B.W. compiled and analyzed the data, and wrote the paper. M.W.I.S.
514 conceived and coordinated the comparative study. Samples were provided by
515 M.P.W.S., M.Ke and M.Kl. Analysis was performed by S.A. (MIR), A.V.M., R.J.S.
516 and W.C.H (NMR), M.Ke and P.S.N. (NEXAFS), L.A.P. and C.A.M.
517 (pycnometry), M.P.W.S. and A.V.M. (BPCA), G.L.B.W. (lipid analysis). All
518 authors provided input into the drafting and the final version of the
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520

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