Synergistic Environmental Degradation of Glass Reinforced Polymer Composites

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Abstract

Synergistic effects involved in the environmental degradation of Glass Reinforced Polymer (GRP) composites were examined in this research. Six GRPs based on E-glass and ECR-glass fibers with four different polymer resins were exposed either individually or in combination to ultraviolet (UV) radiation, water condensation and elevated temperature for approximately 1000 hours. The composites were monitored for weight changes as a function of time and their surfaces were examined after the tests using optical and scanning electron microscopes. A new model of synergistic aging of polymers under UV and water contestation was also proposed. It has been shown that the selected aging conditions created noticeable synergistic effects causing extensive erosion of the polymer matrices of the tested composites which appeared to be much stronger under the combined actions than under individual exposures. The differences in the aging rates under the individual and combined situations were adequately explained using the newly proposed model. It is suggested that the surface erosion of polymer matrices of the GRPs by combined UV and water condensation creates the most effective condition for small polymer particles formed by UV to be subsequently removed by water condensation exposing fresh still undamaged surfaces to further UV degradation. It has also been shown that depending on the size of the particles different particle removal mechanisms occur on polymer surfaces eroded by a cyclic exposure to UV and water condensation.

KEY WORDS: Glass Reinforced Polymers, environmental degradation, synergistic effects ultraviolet degradation, condensation degradation.

1. Introduction

Numerous studies have focused on the environmental degradations of Polymer Matrix Composites (PMCs) in general, and more specifically, aging of Glass Reinforced Polymer (GRP) composites caused by a large variety of aggressive environments. Some of the most recent studies can be found in [1-6]. Very few of them, however, have dealt with the synergistic effects of aging of polymers and GRPs under combinations of several aging conditions [6,7 and 8]. Since this work is concerned with synergistic aging of a group of commercially available GRPs under combined UV radiation, moisture, temperature and time, an emphasis is made here to evaluate the research progress made so far in this very important area of polymer and polymer based composite science.

The UV components of sunlight which reach the ground are in the range of 280-400nm [28]. The energy of ground reaching UV photons is comparable to the dissociation energies of polymer covalent bonds [9-11] resulting in a loss of surface gloss, surface discoloration, chalking, flaking of surface resin, pitting, microcracking, and a severe loss of resin [7-13] in GRPs. The damaging effect of water or moisture on polymer composites, on the other hand, is not as harsh as degradation just by UV radiation even at elevated temperatures [7]. However, moisture diffusion into polymer matrix/fiber interfaces can damage the interfaces by micro-cracking [14, 15] especially at elevated temperatures. In addition, hydrolysis of chemical bonds may lead to permanent chemical degradation [1, 15, 16] and moisture induced swelling [1, 14, 17] of polymers and their composites.

The combined action of both UV and water on polymers and PMCs as a function of time and temperature could be even more severe than the individual effects. Therefore, this topic has been addressed here both experimentally and numerically. It should be stated that the objective of this work was not to rank a group of GRP composites for their resistance to the above aging conditions, but to investigate the fundamental aspects of the combined UV/water aging processes applicable to any GRP, and more generally, to any PMC system.

2. Experimental Procedures

2.1. Tested composites and specimen preparation

The composites tested in this research are listed in Table 1. The exact chemical compositions of the composites were not released for the purpose of this study. Also, the matrix of composite F is proprietary information.

Table 1: Tested GRPs

Label	Matrix	Glass fiber	Weight fraction (%)
A	Vinyl ester Epoxy	ECR-Glass 1	75
B	Vinyl ester Epoxy	E-Glass	75
C	Vinyl ester Epoxy	ECR-Glass 2	80
D	Polyester	ECR-Glass 2	80
E	Epoxy	ECR-Glass 2	80
F	Proprietary	ECR-Glass 2	80

Two groups of unidirectional GRP composites were investigated. In the first group, three composites (A, B and C) with E-glass, ECR-1 glass and ECR-2 glass fibers in a vinyl ester epoxy blend matrix were tested to evaluate the fiber effect on composites' degradation. In the second group, one type of fiber, ECR-2 glass, was considered with four different polymers, namely polyester, vinyl ester, epoxy and a proprietary resin to determine the matrix effect on the degradation processes in composites C, D, E and F. The fibers came from three different glass manufacturers and the composites were supplied by MacLean Power Systems.

1.3 m long, 6 mm thick and 50 mm wide composite plates were made by pultrusion by an outside manufacturer. The weight fractions of fibers in the six systems were very similar (Table 1). The composites were dry cut before testing into 80 mm long specimens with special care to ensure minimum damage during cutting. The cut edges of the samples were sealed with a thin layer of a room temperature vulcanization silicone. Subsequently, the specimens were heat-treated for 72 h in an oven at 80 °C. The other as supplied surfaces of the samples were smooth and glossy before the tests.

2.2. Environmental testing and analysis

A Q-LABQUV/SPRAY/RP chamber was used for individual and combined UV radiation and water condensation aging tests. The UV radiation tests were carried out for 1000 hours with the UV wavelength ranging from 315 to 400 nm at 80°C. An irradiance level of 1.50 W/m² at 340 nm was chosen to more than double the maximum irradiance of natural sunlight at noon. The test temperature was lower than the glass transition temperatures, T_g, of the polymer matrices of the composites, which were higher than 120 °C. The relative humidity (RH) was 3 ± 2% and was recorded by a relative humidity meter placed inside the chamber.

In the water condensation tests the specimens were continuously exposed to deionized (DI) water vapor condensation at 60 °C for 1000 hours with 92± 2% RH. Water condensation was formed by heating deionized water to vapor on a hot panel inside the chamber. In the cyclic UV radiation and water condensation tests, the composites were exposed to alternating cycles consisting of 16 hours of UV at 80 °C followed by 8 hours of water condensation at 60 °C for 1000 hours. The relative humidities were 33 ± 2% and 92 ± 2 %, respectively.

For weight measurements the specimens were removed every 96 hours from the chamber for about 60 minutes and weighed with an Ohaus Voyager Electronic Balance (precision 0.01 mg). The specimens were then randomly repositioned in the chamber to ensure uniform exposure on all surfaces. The percent weight changes as a function of time were calculated using Eq (2.1):

$$\% \text{ weight change} = \frac{W_f - W_i}{W_i} \times 100 \quad (2.1)$$

where W_f and W_i are the final and initial dry weights of the samples at room temperature.

Optical and scanning electron microscopes (SEM, JSM 500 LV) were used to evaluate the surface morphology of the specimens after their exposure. All specimens were prepared and mounted using conventional procedures for SEM observations. Most of the work on the surface deterioration in the composite under the above testing conditions has been reported elsewhere [18].

3. Aging Results and Discussions

3.1 Surface erosion

Surface characteristics of all tested composite specimens after their exposures to UV, water condensation and combined UV/condensation were examined optically and by the SEM. Despite significant differences in their original surface properties such as the amounts of exposed glass fibers, fiber distributions, fiber alignment, fiber diameters, etc. [19], all composites, including A, responded in quite similar fashions to all three test environments with the exception of composite D. Therefore, A was picked as an example to illustrate the aging effects on the composites (Figure 1A). The polyester composite (D) had apparently UV blockers applied to its resin which resulted in less surface damage by UV (Figure 1B). D also had more exposed fibers on the surface (approximately 13%) in comparison with A (2.8%). In some cases the exposed fiber areas were as large as 50% (composite E).

Figure 1. Typical surface characteristics of A and D composites after 1000 hours of exposure to dry hot UV.

Surface damage characteristics under individual UV, water condensation and combined UV and water condensation on Vinyl ester Epoxy ECR-Glass 1 – Composite A are shown in Figure 2 (UV only), Figure 3 (condensation) and Figure 4 (UV/condensation combination). Distinct differences in surface erosion can be observed on the surfaces subjected to the three different aging conditions.

Figure 2. SEM images of exposed surfaces after 1000 hours of UV radiation for Vinyl ester Epoxy ECR-Glass 1 – Composite A; a) low magnification at $\times 300$ and b) high magnification at $\times 1500$.

Figure 2 shows severe matrix degradation with numerous small particles ranging from a micron or smaller to about 20 μm formed on the surface by chalking. This phenomenon was caused by the polymer matrix becoming excessively brittle due to an increased crosslinking or the formation of microcracking due to chain scission resulting from photo-oxidation reactions induced by UV radiation [20]. The removed layer of the matrix was more than 20 μm deep. The fibers and the fiber/matrix interfaces in all composites were unaffected by UV. Previous research reported that degradation on polymers surfaces by UV is a time dependent effect where the photochemical reactions are restricted to the surface of the polymers within a several micron thick layer [21].

Figure 3. SEM images of exposed surfaces after 1000 hours of condensation for Vinyl ester Epoxy ECR-Glass 1 – Composite A.

a) low magnification at $\times 300$ and b) high magnification at $\times 1500$

The micrographs presented in Figure 3 indicate that the matrix and the interfaces of the composite were degraded by water condensation and the amount of degradation was not uniform across the surface. Overall the amount of damage by water for all composites was much less severe than the damage illustrated in Figure 2 for the UV exposure.

Figure 4. SEM images of exposed surfaces after 1000 hours of UV and condensation for Vinyl ester Epoxy ECR- Composite A.

Cyclic exposure to both UV radiation and condensation resulted in severe degradation that was very different from the other two exposures. The damage consisted of extensive matrix erosion and debonding. As shown in Figure 4, the external polymer layer on the specimen surface was completely removed and the underlying glass fibers were exposed. Most importantly, all samples were almost entirely free from any particles larger than a few microns in diameter.

3.2 Mass changes from individual and cyclic exposures

Figure 5 shows weight changes as a function of time for the composites exposed only to UV radiation. The data points for each curve represent averaged measurements obtained from two specimens. It can be observed that the specimens exhibited immediate loss in weight from the very first hours of testing. All GRPs after 1000 hours of exposure lost between 0.03 to 0.056 %

of their weight. The initial decrease in the weight of the specimens can be attributed to the expulsion of volatiles and residual moisture, which occurred in the first 200 hours of testing. Then, continuous “attack” by UV radiation and oxidation at elevated temperature was the main reason for severe matrix degradation and weight losses.

Figure 5. Weight changes for six composites under UV radiation as a function of time.

The effect of water condensation on the weight of the specimens was entirely different in comparison with the UV tests. As shown in Figure 6, all specimens gained weight with time. Again, the data were obtained from two samples for each composite. The specimens gained about 0.04-0.19% by weight after 1000 hours of exposure depending on their chemical composition without reaching saturation.

Figure 6. Weight changes for six composites under water condensation as a function of time.

The combined effect of UV and condensation created another set of weight gain data noticeably different than the individual UV and water condensation exposure. As shown in Figure 7 the specimens started to exhibit a decrease in weight after about 100 hours of exposure after initial small weight gains. The weight losses continued with increasing rates for the remainder of testing. Excluding specimen D, an average of 0.09% decrease in weight was observed after 1000 hours.

Figure 7. Weight changes for six composites under cyclic UV and water condensation as a function of time.

4. Synergistic Aging Effects and Their Model

4.1 Comparison between individual and combined exposures

Despite the fact that the six GRPs investigated in this research were based on three different glass fibers and four different polymer resins with very different surface characteristics, their individual responses to UV, water condensation and the combination of UV and condensation were quite similar. As expected, all of them lost weight under UV exposure (Figure 5). Also, as expected, all of them gained weight when exposed to water condensation (Figure 6). Under the combined action of both UV and condensation, all of them lost weight (Figure 7). The changes in weight within the 1000 hours of testing were not large but significant enough to detect the effect of the three different environments on the composites. If the sums of the weight changes under the individual exposures are compared with the changes in weight under the combined UV/condensation condition, a very interesting effect can be observed immediately (Figure 8).

The sums of the individual effects on weight changes are positive and much smaller than the changes in weight under the combined environments, which are not only larger but also negative. This clearly demonstrates that the synergistic effects in the aging under UV/condensation/temperature/time of the six composites are quite strong.

The synergistic effects observed in Figure 8 are even better visible in the averaged weight changes obtained from the six composites and presented in Figure 9. Several important observations can be made here. First, the averages of the individual effects on the weight changes for the six composites are positive and about 50% smaller than the averages in the weight changes from the cyclic exposures which are negative. Most importantly, the sum vs. the cyclic average of all composite relations with temperature is very similar to the relations for the individual composites. Therefore, it is possible that the observed synergistic effect is a typical feature for any unidirectional GRP composite exposed to the aging conditions of this research.

Second, the cyclic UV and water condensation resulted in the largest weight reductions, and most likely the largest amounts of damage to all six composites in comparison with the other exposures. Third, the effect of UV on the averages under UV only seems to increase with a decelerating rate. On the other hand, the cyclic effect of UV and water condensation appears to accelerate. Finally, the standard deviations of the averages from the UV tests are the smallest whereas the condensation tests resulted in the largest scatters in weight charges. Regarding the scatters, the averaged sums and averaged cyclic data for all composites are much closer to the condensation scatters than the UV scatters. Condensation alone or in combination with UV seems to create a higher level of uncertainties in the response of the composites to aging. Most likely this is caused by a much stronger and less predictable effect of interfaces in the condensation erosion than in the UV degradation, which seems to be more predictable and dominated by the steady erosion of the matrix between the fibers.

Figure 8. Weight changes for six composites tested under cyclic UV/condensation and the sums of weight changes under individual UV and condensation conditions with time.

Figure 9. Averaged weight changes with standard deviations for six composites tested under individual UV and water condensation, cyclic UV and condensation, and the averaged sums of weight changes from the individual exposures as a function of time.

Two main mechanisms should be considered to explain why the surfaces of the composites after the cyclic UV/water condensation exposures are severely eroded but almost free from any visible particles (Figure 4). One is based on the chemical degradation of the resins under high energy radiation conditions with oxygen, for example by the formation of hydroperoxides, a significant intermediate [11, 20]. Those instable intermediates are quite active and result in chain scission and crosslinking after complex reactions. Eventually, these chemical reactions will result in the microcracking and chalking on the exposed surface, which was shown in Figure 2. During the

condensation cycle, resin hydrolysis and moisture absorption occur [20] but the resin erosion is significantly less severe [7]. In the second step, any soluble and insoluble (particles) products are removed from the ultraviolet radiated surface by slowly moving condensation water. Then a fresh polymer surface is exposed to the following UV cycle and more damage is subsequently created by UV followed by the subsequent cleaning of the surface damage by water in the next condensation cycle.

4.2 Model of synergistic polymer degradation

In support of the above two mechanisms of UV degradation and subsequent removal of UV generated particles by water condensation, a synergistic model of aging has been proposed in this work with the following assumptions:

- Constant air temperature, humidity and UV radiation intensity.
- Contributions from individual wavelengths are independent.
- Only a fraction of the absorbed photons lead to photolytic activity. The photo recovery effect is neglected.
- The initial polymer surface exposed to UV has a sinusoidal shape.
- No water absorption or diffusion occur during condensation exposure.
- Water does not react with polymer.
- No thermal degradation takes place.
- Particles are spherical in shape and the local surface imperfections do not affect the particle removal process.
- UV light is from a line source. When a sinusoidal surface is exposed to radiation, UV intensity is a function of the angle between the surface and the incident light.
- Water flow is laminar and is caused by gravity alone.

4.2.1. UV radiation model

Since UV irradiation is the initiator for polymer degradation, an approximate solution is obtained

for the total effective UV irradiation dosage based on Eq. (4.1) [9, 22, 23]

$$D_{total(t)} = \int_0^t \int_{\lambda_{min}}^{\lambda_{max}} E_o(\lambda, t) (1 - e^{-A(\lambda, t)}) \phi(\lambda) d\lambda dt \quad (4.1)$$

where

- λ_{min} : minimum photolytically effective wavelengths (nm)
- λ_{max} : maximum photolytically effective wavelengths (nm),
- $A(\lambda, t)$: the absorbance of the light of wavelength λ in the material and at time t , (dimensionless),
- $E_o(\lambda, t)$: incident spectral UV radiation dose to which a polymeric material is exposed to at time t ($J\ cm^{-2}$),
- t : elapsed time (second).
- $D_{total}(t)$: total effective dosage (J)
- $\Phi(\lambda)$: quantum yield, which is the number of times a specific event occurs per photon absorbed by the material.

A linear function is then used to establish a relationship between the total irradiation dosage after a certain amount of time, D_{total} , and the amount of UV damage defined by the local damage thickness, Γ , [23]

$$\Gamma = a \times D_{tot} \quad (4.2)$$

where a is a factor indicating the efficiency with which the UV degraded material is removed from the surface during UV exposure only.

The intensity of irradiation for any angle of UV light with respect to the surface is determined by

$$E_o = I_o(\lambda, \theta, t) \cos(\theta) \quad (4.3)$$

where I_o is the intensity of UV ($W/m^2 \cdot s$), and θ is the angle between the UV light and the normal direction to the surface.

The numerical differentiation method was adopted to solve Equations 4.1-4.3 in the simulations of UV degradation of polymer surfaces with various sinusoidal profiles. All physical input data used in the simulations are listed in Table 2. If we assume that (1) all UV generated particles formed on the degraded surface are immediately removed after their formation so a “fresh” surface is subsequently exposed to UV, (2) a parameter “ a ” is selected, and (3) the incident UV light shines perpendicularly to the horizontal direction, the stages of surface reconfiguration by UV of a standard sinusoidal surface with the arithmetical mean roughness of 0.466 can be observed in Figure 10. For the selected light direction there are the maxima of degradation rates at the peaks and the valleys of the surface where the degradation rates are faster than those at the other locations.

Table 2: Input data for performed UV damage simulations.

Parameter	Value
Wavelength	340 nm
Radiation intensity, I_0	1.5 W/m^2
Quantum yield Φ - [24]	10^{-4}
Exposure time t	1000 hours
Exposure area	$3.25 \times 10^{-3} \text{ m}^2$
Efficiency of material removal; arbitrarily chosen	0.1

Figure 10. Stages of UV degradation of a sinusoidal surface.

4.2.2. Hydrodynamic and particle removal models

After UV exposure for a certain period of time there will be a number of small particles of different shapes and sizes formed on the polymer surface (see Figure 2). These particles will be subsequently washed away by the drag force of the water flow in the condensation cycle. The drag is caused by the viscous shear stresses exerted on the solid-fluid interface. The simulation results of the water velocity and shear stress distribution on the polymer surface are shown in Figure 11 based on the commercial finite element software COMSOL, using the physical data listed in Table 3 and for the same initial sinusoidal surface as assumed in Figure 10. The open boundary condition was assumed on the water surface. It can be seen in Figure 11 that there are periodic oscillations in the viscous stresses on the polymer surface caused by the initial surface configuration. The decrease in the shear stress in Figure 11b was caused by an increase in the velocity of the water flow from an initial value of 10 mm/sec to 340 mm/sec at the end of the specimen.

Table 3. Input data used in the hydrodynamic model.

Parameter	Value
Initial water flow rate	$1 \times 10^{-2} [\text{m/s}]$
Thickness of water layer	0.5mm
Surface amplitude	0.02 mm
Dynamic viscosity of water at 25° C	$8.94 \times 10^{-4} \text{ N}\cdot\text{s}/\text{m}^2$
Density	$1000 \text{ kg}/\text{m}^3$
Body force (gravity)	$9.8 \text{ m}/\text{s}^2$

Figure 11. (a) Velocity of water flow; (b) Viscous stress along the surface.

To determine whether the UV generated particles can be removed by water flow, the drag force has been compared with the Coulomb friction between the particles and the solid surface. For micro-scale particles, the effects of gravity and buoyancy are relatively small and can usually be neglected [25]. As a result, the friction is proportional to the normal stress acting on the surface, which is solely determined by the van der Waals forces. For simplicity, the particles are assumed to have spherical shapes for estimating the van der Waals forces. In addition, three different particle removal mechanisms namely lifting, sliding and rolling have been considered in the model [25, 26, 27].

Figure 12. SEM images of composites E after (a) UV radiation and (b) UV & water condensation.

The Johnson-Kendall-Roberts (JKR) model [26] has been used to calculate the van der Waals forces. By combining an average shear stress of 1.59 N/m^2 from our hydrodynamic model, contact mechanics [26] and particle detachment equations [25,26,27], preliminary results have been obtained showing that particles of diameter larger than $1.3 \text{ } \mu\text{m}$ can roll, and that the sliding type of motion is possible for particles of diameters $34.2 \text{ } \mu\text{m}$ and above. Particles larger than $84.6 \text{ } \mu\text{m}$ will be lifted by water. In other words, particles larger than $1.3 \text{ } \mu\text{m}$ in diameter will be removed from the surface by three different mechanisms, whereas those smaller than $1.3 \text{ } \mu\text{m}$ will adhere to the surface.

In order to validate the predictions, more SEM imaging was performed on the ECR 1/epoxy composite (Figure 12) after the combined UV/water condensation exposure. The results showed

that a slow water flow during condensation removed all larger particles (2-10 μm) after UV radiation; however, some smaller particles ($<1\ \mu\text{m}$) were still adhered to the surface, as seen in Figure 12b. Similar observations can be made from the surface characteristics presented in Figures 4. After the removal of the larger particles by water condensation, a “fresh” surface of the polymer matrix is exposed to UV again, and the process continues with a slowly increasing rate (Figures 8 and 9).

Conclusions

It has been shown in this study that commercially available unidirectional GRP composites when exposed to individual and cyclic UV radiation, water condensation and temperature develop significantly different degradation mechanisms with strong synergistic effects. Most importantly the sums of the individual UV and water condensation aging effects on weight changes are positive with weight gains, whereas the weight changes under the cyclic combined conditions are negative with weight losses for all six tested composites. The average weight losses under cyclic conditions for all six composites were found to be about 100% larger than the weight gains from the simple superposition of the UV and condensation environments for the same composites.

To explain the synergistic effect observed in this work, a new hypothesis of damage initiation and progression in the GRPs subjected to both UV and water condensation was formulated and supported by a new model of synergistic aging. According to our model, under the combined action of UV and water condensation small particles are first formed on the surfaces of polymeric surfaces exposed to UV radiation. The particles are subsequently removed by the slowly moving water caused by the condensation, which exposes fresh undamaged polymer surfaces to the cycle of UV. This creates a much more severe surface erosion than UV alone. Using our still very preliminary model, UV generated polymer particles found on the surfaces of the tested GRPs after UV were evaluated for their response to water cleaning during the condensation cycle. It was determined that most of the particles generated during the combined testing should be removed by water condensation. This prediction was successfully verified by SEM observations of the environmentally aged composites.

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