2D-Imaging of Sampling-Probe Perturbations in Laminar Premixed Flames using Kr X-ray Fluorescence

N. Hansen,^{1} R.S. Tranter,² K. Moshhammer,^{1,3} J.B. Randazzo,² J.P.A. Lockhart,² P. G. Fugazzi,¹
T. Tao,^{1,4} A.L. Kastengren⁵*

¹Combustion Research Facility, Sandia National Laboratories, Livermore, CA 94551, USA

²Chemical Sciences and Engineering Division, Argonne National Laboratory, Argonne, IL
60439, USA

³Physikalisch Technische Bundesanstalt (PTB), Bundesallee 100, 38116 Braunschweig,
Germany

⁴Center for Combustion Energy and Department of Thermal Engineering, Tsinghua University,
Beijing 100084, PR China

⁵X-Ray Sciences Division, Argonne National Laboratory, Argonne, IL 60439, USA

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* Corresponding author: email: nhansen@sandia.gov; phone: +1 925 294 6272; fax: +1 925 294 2276

Abstract

The perturbation of the temperature field caused by a quartz sampling probe has been investigated in a fuel-rich low-pressure premixed ethylene/oxygen/argon/krypton flame using X-ray fluorescence. The experiments were performed at the 7-BM beamline at the Advanced Photon Source (APS) at the Argonne National Laboratory where a continuous beam of X-rays at 15 keV was used to excite krypton atoms that were added to the unburnt flame gases in a concentration of 5% (by volume). The resulting krypton X-ray fluorescence at 12.65 keV was collected and the spatially resolved signal was subsequently converted into the local temperature of the imaged spot. One and two dimensional scans of the temperature field were obtained by translating the entire flame chamber through a pre-programmed sequence of positions on high precision translation stages and measuring the X-ray fluorescence at each location. Multiple measurements were performed at various separations between the burner surface and probe tip, representing sampling positions from the preheat, reaction, and postflame zones of the low-pressure flame. Distortions of up to 1000 K of the burner-probe centerline flame temperature were found with the tip of the probe in the preheat zone and distortions of up to 500 K were observed with it in the reaction and postflame zones. Furthermore, perturbations of the temperature field have been revealed that reach radially as far as 20 mm from the burner-probe centerline and about 3 mm in front of the probe tip. These results clearly reveal the limitations of one-dimensional models for predicting flame-sampling experiments and comments are made with regard to model developments and validations based on quantitative speciation data from low-pressure flames obtained via intrusive sampling techniques.

Keywords:

low-pressure flame; speciation data, probe perturbation; X-ray fluorescence; chemical models

Introduction

Laminar premixed flames are widely used in combustion chemistry research [1]. Especially under sub-atmospheric pressure and in combination with flame-sampling molecular-beam mass spectrometry (MBMS), this flame configuration has been considered to be ideal for studying the chemical details of fuel-specific combustion processes [2, 3]. In a typical laminar premixed low-pressure flame experiment, gases are sampled through a sonic quartz nozzle from within the flame, quenched by expansion and subsequently analyzed using mass spectrometry. These data can then be converted into spatially resolved species' mole fractions as a function of distance from the burner surface when gases are sampled from multiple flame positions [1]. Pioneered by Biordi and co-workers [4, 5], flame-sampling MBMS has revealed the identity of many important combustion intermediates and provided valuable targets for the development and validation of detailed chemical models. Many chemical pathways and reaction mechanisms in flames have been identified and established by comparing the experimental results from the flame structure studies with results from model simulations [1-3, 6].

However, these flame-sampling experiments are invasive, and the sampling nozzle affects the physical and chemical structure of the flame. A description of these effects has not yet been added onto the normally used one-dimensional, laminar flame models (e.g. PREMIX in CHEMKIN [7]), and care should be taken when comparing experimental results with model calculations [1]. In order to provide a more meaningful interpretation of the experimental and modeling results, a comprehensive physical and chemical description of the probe-induced perturbations needs to be implemented in both models and data analysis procedures.

The current understanding of the effect that a sampling probe, which is typically made out of quartz, has on a flame is that it acts as a heat sink, distorts the temperature profile and the

flow field, and hence alters the species' mole fraction profiles relative to those in an unperturbed flame [1, 8, 9]. All of these effects are linked to one another; however, given the complexity of this issue, an overarching analysis of these effects has not yet been provided. Alongside theoretical considerations [10-13], experimental work has been focused on the analysis of the flow dynamics [14-16], the comparison of species profiles derived via MBMS and laser-based techniques [17-21], and the determination of composition [22] and temperature effects [8, 21, 23-26].

One of the most important components of using models to obtain a thorough interpretation of the experimental observations is a precise and accurate knowledge of the temperature history a molecule has experienced. Consequently, various attempts have been made to measure the flame temperature in “perturbed” and “unperturbed” flames. These measurements, mainly using laser-based diagnostic and thermocouples, along the centerline of the flame revealed differences between the “perturbed” and “unperturbed” flames of around 100 to 500 K in the postflame zone [8, 21, 23-26]. In the preheat and reaction zones, the temperature gradient was found to be less pronounced in the probe-distorted flame, resulting in an apparent shift of the “perturbed” temperature profile further away from the burner. While Bastin *et al.* [23] used the entire “perturbed” profile as an input to model calculations, Douté *et al.* [24], Hartlieb *et al.* [25], Vovelle *et al.* [26], and Struckmeier *et al.* [8] argued that sampling position-sensitive corrected temperature profiles best represented the temperature history an individual non-reacting atom would experience when flowing from the burner to the nozzle.

Two-dimensional probe effects on the temperature field were qualitatively visualized using NO laser-induced fluorescence by Struckmeier *et al.* [8] who concluded that temperature gradients in the radial direction may be quite considerable and should not be ignored. Similar

deviations from the one-dimensionality of the temperature field were also observed in simulations carried out by Skovorodko *et al.* [11], Gururajan *et al.* [13], and Deng *et al.* [10].

To develop a better quantitative understanding of the two-dimensional effects of the sampling probe on the temperature field, we performed synchrotron-based X-ray fluorescence (XRF) spectroscopy measurements of the gas density, from which the temperature can be calculated, around the quartz sampling probe using krypton as the fluorescent species. Highly spatially resolved raster scans of gas densities around the probe were obtained for multiple distances of the probe tip to the burner surface. The data reveal the magnitude of the two-dimensional distortions of the temperature field caused by a sampling probe compared to the unperturbed flame at an unprecedented level of detail and accuracy.

The experimental results presented in this study are intended to ultimately allow for a minimization of experimental uncertainties in flame-sampling experiments. The long term goals of these efforts are to provide guidance for probe design and empirical corrections for specific probe geometries. Ultimately, this work should result in validation targets for the detailed chemical models of the highest quality and accurate models for chemically detailed simulations. Furthermore, the results highlight the need to address two-dimensional modeling approaches for sampling results from low-pressure flame experiments.

Experimental Procedures

The experiment consists of a low-pressure flame chamber in which a vertically translatable stainless-steel McKenna-type burner (6 cm diameter) and the quartz sampling probe are mounted. The flame we used to study the probe perturbation was a fuel-rich (stoichiometry of 1.7) ethylene (C_2H_4)/ O_2 /Ar/Kr flame with a cold gas composition by volume of 14.5% C_2H_4 ,

25.5% O₂, 55% Ar, and 5% Kr at a total flow rate of 4 slm. The Kr concentration was selected to produce sufficient fluorescence signal while at the same time keeping the overall Kr usage reasonable. A shroud gas of Ar was supplied at a rate of 4 slm. The pressure in the flame chamber was kept constant at $p = 30$ Torr with a combination of a butterfly valve, a baratron, and a mechanical oil pump. The region behind the quartz probe was pumped by a turbo molecular pump. The design of the sampling probe is similar to those used in flame sampling studies by Hansen and co-workers [2] and is shown in the Supplementary Material (S1). All the experiments reported here used the same probe which had an opening in the tip of ~ 250 μm in diameter.

The experiments were performed at the 7-BM beamline of the Advanced Photon Source (APS) at the Argonne National Laboratory [27]. A schematic of the experimental setup is shown in Fig. 1.

Details of the Kr-XRF set-up are described in Ref. [28] and only a few details are provided here. The beamline produced a nearly collimated X-ray beam of monochromatic X-rays (15 keV mean photon energy, 1.2% spectral bandpass, $\sim 1.6 \times 10^{11}$ photons/s incident flux), which was focused with a pair of 300 mm long Kirkpatrick-Baez focusing mirrors to a $5 \mu\text{m} \times 7 \mu\text{m}$ (full width at half maximum) focus. The flame chamber was positioned so that the sampling probe was at the same z position as the beam focus. The position of the X-ray focal point at the tip of the quartz probe was defined as $x = y = z = 0$ mm, with the x -, y -, and z -axes defined as shown in Fig. 1. The beam entered and exited the flame chamber along the z -axis via thin (125 μm) polyimide windows. A diamond photodiode monitored the incident beam intensity prior to the flame chamber and a silicon PIN photodiode after the flame chamber was used to determine the amount of absorption, particularly for locating the sampling probe in the chamber.

Due to the absorption of an incident X-ray photon, the Kr in the flame is ionized, predominantly through the ejection of a K-shell (1s) electron (14.326 keV K-edge energy [29]). The resulting ion subsequently relaxes either through emission of an Auger electron or through decay of a 2p electron to the 1s level, resulting in the emission of an X-ray fluorescence photon with 12.65 keV energy (Kr $K\alpha$) [30]. These Kr fluorescence photons were collected orthogonal to the incident beam with a polycapillary X-ray optic (100 mm nominal focal length, XOS) and detected using an energy-dispersive X-ray detector (silicon drift diode, SII-Nano Vortex EM-60). The polycapillary optic only allowed X-rays from a limited region (300 μm FWHM) of the beam path (along the z-axis) to reach the detector. As such, the measurement probes a discrete volume 5 x 7 x 300 μm FWHM in size, with a technique known as X-ray confocal microscopy [31].

One-dimensional horizontal (x) and vertical (y) scans, as well as two-dimensional raster scans, were accomplished by maneuvering the flame chamber on high-precision (better than 5 μm) translation stages around the X-ray beam while keeping the polycapillary optic and fluorescence detector fixed. The Kr-XRF signal was recorded from up to -20 to +2 mm in the x direction and -0.5 mm to HAB mm in the y direction, where HAB (height above burner) represents the multiple burner-cone separations probed. For the current experiments, step sizes of 200 to 250 μm were found to be an adequate compromise between spatial resolution and the time needed to finish the pre-programmed one- and two-dimensional scans. Because of the assumed symmetry of the probe perturbation, scans along the z-axis were not found necessary and as such, the spatial resolution of the experimental horizontal (x) and vertical (y) scans described here is only controlled by the step sizes of 200 to 250 μm given the high-precision of the translation stages and the size of the X-ray focal point.

The collected fluorescence signal can be used as a measure of the gas density because the total X-ray fluorescence emission is directly proportional to the flux of the incident X-ray beam (ϕ), as measured by the diamond photodiode [see Fig. 1], and the molecular density of krypton in the imaged focal point. The photon flux ϕ is an independent parameter related only to the performances of the synchrotron and the X-ray optics. It has been maximized to result in the strongest fluorescence signal S , while ensuring a linear correlation between S and the concentration of the fluorescing Kr atoms. Signal trapping of the fluorescence signal is anticipated to be negligible under the current conditions and the detected fluorescence signal S is correlated to the temperature via

$$S(x,y) \sim \phi \times x_{\text{Kr}}(x,y) \times n/V \sim \phi \times x_{\text{Kr}}(x,y) \times p/T(x,y) \quad (1)$$

where x and y are the coordinates as introduced in Fig. 1, $T(x,y)$ is the local temperature, $x_{\text{Kr}}(x,y)$ is the mole fraction of krypton at the respective flame position, and n/V is the number density that can be expressed as $p/T(x,y)$ from the ideal gas law which is appropriate for the conditions studied. Eq. (1) is based on correlations that are well-known from fluorescence spectroscopy [32] and the validity of Eq. (1) is shown with the calibration and cold-gas measurements described below. For these burner-stabilized low-pressure premixed flames the pressure p can be considered to be constant. The fluorescence signal was typically collected for 30 s/point which resulted in point-to-point signal fluctuations that, after correction for the incident X-ray beam intensity, correlate to an accuracy of the temperature measurements of ± 50 K in the postflame zone. Because of the lower temperature and larger x_{Kr} in the preheat zone (i.e. stronger fluorescence signal), the accuracy of the temperature measurements in the preheat and reaction zone is in the order of ± 20 K.

The absolute fluorescence signal was calibrated at $p = 30$ Torr against cold-gas (298 K) Kr-Ar mixtures of known mole fractions between $x_{\text{Kr}} = 0$ and 0.08; a linear response of the fluorescence signal as a function of the concentration was observed (see Supplementary Material S2). The determined calibration factor (c) was then used to calculate the flame temperatures at a constant pressure using the following equations:

$$S(x,y) = c \times \phi \times x_{\text{Kr}}(x,y) \times 1/T(x,y) \quad (2)$$

or
$$T(x,y) = c \times \phi \times x_{\text{Kr}}(x,y) \times 1/S(x,y) \quad (3)$$

As seen from Eq. (3), two-dimensional measurements of x_{Kr} are required to determine the local flame temperatures $T(x,y)$. However, this information is not currently experimentally accessible and thus two approximations were made: In a first approximation, we measured $x_{\text{Kr}}(x,y)$ along the center line of the flame ($x = 0$ mm) in separate experiments at Sandia National Laboratories by electron-ionization flame-sampling molecular-beam mass spectrometry using the instrument and routines described elsewhere [33, 34]. Despite the two-dimensional temperature distortions revealed in this paper, this approximation seems warranted because flame-sampling mass spectrometry has been proven multiple times to retrieve major species mole fraction profiles that, within well-established uncertainty limits, agree with model calculations [2-4]. For this experiment, we used a 14.5% C_2H_4 /25.5% O_2 /60% Ar flame at $p = 30$ Torr and then assumed x_{Kr} of the flame at the APS to be $1/12^{\text{th}}$ of x_{Ar} from the MBMS measurements. This assumption is reasonable because neither of the rare gases Kr and Ar participates in the combustion process and the uncertainties introduced due to differences in their diffusion processes are negligible compared to the experimental uncertainties. The measured $x_{\text{Kr}}(x=0,y)$ profile is provided in the Supplementary Material (S3). Overall, the XRF experimental signal

and analysis procedure result in a combined temperature uncertainties of ± 150 K in the postflame and ± 50 K in the preheat and reaction zone.

As shown below, this approximation was validated by the agreement of the XRF results with independent approaches for the unperturbed flame [temperature measurement via laser-induced fluorescence (LIF) of the OH radical] and the perturbed flame (temperature profile obtained via molecular-beam sampling data). The unperturbed centerline temperature profiles were measured with laser-induced fluorescence using the frequency-doubled output of an optical parametric oscillator (Continuum Sunlite EX OPO) near 306 nm to excite the OH A-X (0,0) transition as described in Ref. [35]. The accuracy of these independent measurements is estimated to be ± 150 K.

In a second approximation, we assumed $x_{Kr}(x,y)$ to be constant along the x-axis at any given distance to the burner surface (y value). Given the two-dimensional temperature distortions reported in this work, this assumption is not entirely accurate and concentration gradients might exist along the x-axis. As mentioned above, the two-dimensional $x_{Kr}(x,y)$ is experimentally not accessible and at present, a complete two-dimensional simulation would be required to determine $x_{Kr}(x,y)$. However, such multi-dimensional model simulations would also require $T(x,y)$ as input parameters and are outside the scope of the present work. Consequently, the results presented below, which are based on the above approximation, likely underestimate the temperature distortions somewhat and should be considered a lower limit.

Results and Discussion

This section is organized as follows: First, we present data from unburnt gas mixtures with known compositions to check for the applicability and validity of the experimental method.

Second, we present two-dimensional raster scan images and one-dimensional vertical scans of the unperturbed flame that were taken in the absence of the sampling probe. These data serve as a reference for the measurements described in the third part, in which we discuss the two-dimensional flame perturbations at multiple burner-probe separations.

a) Cold-gas Experiments

To check for the proper performance of the experimental set-up and to calibrate the fluorescence signal, we measured the Kr X-ray fluorescence from gas mixtures with known compositions. For the results shown below, Ar and Kr at flowrates of 3850 sccm (Ar) and 200 sccm (Kr) were mixed upstream of the water-cooled McKenna-type burner through which the mixture flowed. The pressure in the flame chamber was kept constant at $p = 30$ Torr. For these experimental conditions Eq. (1) indicates that the detectable fluorescence signal $S(x,y)$ at any given position should be constant in the entire scanned area between the burner surface and the quartz probe tip, assuming x_{Kr} , p and T remain constant. The assumption of a constant x_{Kr} seems to be warranted given that the cold-gas convection velocity (69.9 cm/s), which is the same for both components, is much faster than the diffusional velocity in this two-component mixture.

The experimental results are shown in Fig. 2(a) as a two-dimensional image around the quartz probe (with the tip positioned at 8.4 mm above the burner surface) and in Figs. 2(b) and (c) as horizontal and vertical traces. The homogeneity observed in the two-dimensional Kr X-ray fluorescence image and the flatness of the fluorescence signal in the horizontal and vertical scans shows the suitability of the present approach to measure gas-densities in low-pressure

experiments. It also demonstrates that there is no influence from the nozzle on the temperature of the cold gas mixture.

b) Unperturbed Flame

Measurements of the Kr XRF were performed in the unperturbed ethylene flame by replacing the quartz sampling probe with a quartz disk. A two-dimensional raster scan was recorded from $x = -12.0$ to 0.0 mm and $y = 23.5$ to 30.0 mm above the burner surface. The result of the two-dimensional image is shown in Fig. 3(a). As can be seen a homogenous looking image was recorded, in which the signal intensity is shown with darker colors indicating smaller signal intensities. Horizontal scans from $x = -12.0$ to 0.0 mm at 34.0 mm and 19.0 mm above the burner surface revealed the one-dimensionality of the temperature field in the probed region. In addition, vertical one-dimensional scans were recorded at $x = 0, 5.0, 12.0, 15.0, 20.0$ and 25.0 mm away from the burner-probe centerline.

According to Eq. (3) the XRF signal can be converted into temperatures when the mole fraction x_{Kr} as a function of the x and y coordinates and the calibration factor (as determined at 298 K) are known. As described above, we used the $x_{Kr}(x = 0 \text{ mm}, y = 0 \text{ to } 30.0 \text{ mm})$ from the EI-MBMS measurement (along the centerline over the burner surface) and the cold-gas (298 K) calibration data to convert the raw data into temperature profiles. The results are shown in Fig. 3(b). Fluorescence signals taken along the centerline ($x = 0 \text{ mm}, y = 0 \text{ to } 30.0 \text{ mm}$) and 15.0 mm away from the centerline ($x = 15.0 \text{ mm}, y = 0\text{-}30.0 \text{ mm}$) result in nearly identical temperature profiles, indicating the flatness of the flame. The results reveal temperatures of 750 K at the burner surface and 2100 K in the exhaust region which are reasonable temperatures at these HABs for this flame. However, the vertical scans at $x = 20.0$ (not shown) and 25.0 mm [shown in Fig. 3(b)] reveal some deviations from the one-dimensionality of the low-pressure

premixed “flat” flame. Such deviations became visible as a larger temperature in the postflame region. A likely explanation is that smaller signal intensities are observed towards the postflame zone due to shroud gas diffusing into the flame, thus resulting in a lower krypton mole fraction than at the centerline x_{Kr} .

The temperature along the centerline was additionally measured with OH LIF as described above and the results are included in Figure 3(b). The good agreement between XRF and LIF measurements can be interpreted as validation of our data analysis procedures. In summary, the Kr XRF measurements in the unperturbed $C_2H_4/O_2/Ar/Kr$ flame allowed for determining the flame’s unperturbed temperature profile. In the next step, we used this unperturbed profile in combination with the Kr fluorescence measurements in the sampling-probe perturbed flame to quantify the probe-induced temperature distortion.

c) Sampling-Probe Induced Perturbed Flame

To measure the temperature drop around the quartz probe we performed scans of the Kr fluorescence from the probe-perturbed flame. An example of the experimental results is given in Fig. 4. For this experiment, the probe burner separation was chosen to be 4.2 mm so the probe tip was in the reaction zone of the flame and the two-dimensional range from $x = -12.0$ to 0.0 mm and $y = 4.7$ to 1.5 mm was scanned. As in Fig. 3(a), the signal intensities are shown as a different color, with darker colors indicating lower signal intensities. The sampling probe is located in the top right corner. It can easily be seen that unlike Fig 3(a) the temperature field is no longer homogeneous in the radial direction.

Following the same procedures as applied for the unperturbed flame and as outlined in the “Experimental Procedures” Section, the raw XRF signals were converted into local flame temperatures. Two-dimensional images of the probe-perturbed temperature field are shown in Figs. 5(a) and (b) for burner-probe distances of 4.2 mm (reaction zone) and 2.0 mm (preheat zone), respectively. Isotherms have been added to the images to guide the readers’ eyes. Large two-dimensional perturbations are immediately visible. Conversely the temperatures measured in the unperturbed flame, Fig. 3(b), varied only in the vertical direction and isotherms ran horizontally (parallel to the burner surface) until edge effects perturbed the flow. Vertical and horizontal traces from such images of the probe-perturbed temperature fields will be discussed in the following paragraphs.

First, we discuss the horizontal temperature scans from $x = -12.0$ to $+2.0$ mm at a burner-probe separation of 4.2 mm. The results are shown in Fig. 6 for four different distances of the X-ray beam to the probe tip.

Figure 6 indicates that the temperature changes from the front of the nozzle (radial distance $x = 0$ mm) to the outside (unperturbed) regions by ~ 700 K with the X-ray beam 0.2 mm below the nozzle. The temperature change becomes smaller for larger distances away from the quartz tip, but even at a distance from 2.7 mm (1.5 mm above the burner surface), the temperature drop is still around 400 K. These results are not unexpected because similarly large numbers in the temperature drop were observed in fuel-rich propene/O₂/Ar and *n*-heptane/O₂/Ar flames [24, 25]. More interestingly, Figs. 5 and 6 reveal significant distortions of the temperature along the radial x -axis. It can be seen that perturbations reach as far as 8 to 10 mm before the temperatures of the unperturbed flame are reached. Even larger distortions of the temperature field were observed between perturbed and unperturbed flames when “sampling” from the

preheat zone [Figure 5(b)]. Here, deviations from the unperturbed flame were revealed that reach as high as 900 K when comparing the temperature in front of the nozzle to the unperturbed flame temperature shown in Fig. 3(b) at 2 mm. From the non-horizontal behavior of the isotherms shown in Fig. 5(b) over the scanned region, it is obvious that the perturbations extend beyond 12.0 mm away from the centerline (see below).

Based on these results and to show the effect of the burner-probe separation on the temperature distortions, we have performed horizontal fluorescence scans for burner-probe separations of 2.4, 4.2, and 31.0 mm, representing “sampling” from the preheat, reaction, and postflame zones. For this, scans were recorded at 0.2 mm below the tip of the probe in the range from $x = -24.0$ to $+2.0$ mm for the 2.4 mm separation and -12 to $+2$ mm for the 4.2 and 31.0 mm probe-burner separation. The results are shown graphically in Fig. 7. The solid lines present a Gaussian curve fit through the data. It can be seen that when sampling at 0.2 mm below the quartz probe tip, the width of the probe-perturbed range along the radial axis is different at the different burner probe separations. While at 4.2 mm the flame seems “unperturbed” around 10 mm away from the probe tip, the distortion is much wider when the probe tip is in the preheat region of the flame at 2.4 mm. In the postflame zone, the distortion appears somewhat narrower. Again, please also note the temperature drop of ~ 800 K in front of the probe tip when sampling in the preheat zone of the flame and the radial distortion extending ~ 18 mm from the centerline.

The observed radial distortions of the temperatures seem bothersome, considering that the overall burner radius is 30 mm and edge effects were seen as close as 20 mm from the centerline, resulting only in about 10 mm of an “unperturbed” flame. The observed horizontal temperature gradients are likely to result in concentration differences and consequently diffusion processes along the x-axis, thus supporting simulations performed by Skovorodko *et al.* [11] which

revealed a significant effect of radial diffusion on the distribution of concentrations in the plane of the orifice. These results obviously indicate limitations of the currently used assumption of one-dimensionality for sampling results from low-pressure flames, as it is not obvious what temperature history the sampled molecules have experienced.

Because of the widespread use of one-dimensional flame models, vertical perturbed temperature scans (along the y-axis) on the centerline ($x = 0$ mm) of the perturbed flame are described next. While the interpretation of the horizontal temperature scans (along the x-axis) was based on the assumption of a constant x_{Kr} as discussed earlier, the interpretation of the vertical scans is more complicated because x_{Kr} changes as function of the y-coordinate. Similarly to the analysis of the unperturbed flame data, the analysis of the vertical temperature scans in the perturbed flame is based on x_{Kr} along the centerline as determined by molecular-beam mass spectrometry.

The results for the probe-distorted temperature profiles on the centerline are shown in Fig. 8 for burner-probe separations of 2.0, 4.2, 7.1, 15.9, 23.5 and 31.0 mm. For the positions where the probe is in the postflame region, the temperature along the centerline as function of distance from the burner surface coincides with the unperturbed temperature profile when measurements are made at least 3-4 mm below the probe tip. In this range, the temperature steadily decreases and deviates from the unperturbed profile by up to ~500 K as seen in the previously described horizontal scans. Such long range axial perturbations were also observed theoretically by Gururajan *et al.* [13] and experimentally by Hartlieb *et al.* [25] and Struckmeier *et al.* [8]. The latter two studies revealed extreme probe perturbations to the temperature profile that extend 10-15 mm along the centerline below the probe tip. In this work the XRF technique has allowed much more detailed temperature profiles to be obtained with better spatial

resolution. As a result, it can be concluded that the perturbations caused by the sampling probe, when positioned in the reaction and postflame zone of this fuel-rich flame, expand over a range of at least ± 10 mm (x-axis) $\times$ 3 mm (y-axis) for the flame conditions and the current sampling probe shape.

From the results shown in Fig. 8, it is also obvious that at burner-probe distances representing “sampling” from the preheat and reaction zone the temperature profiles are significantly influenced by the sampling probe, resulting in perturbed temperatures that are below the unperturbed temperatures even at the burner surface. These low (perturbed) temperatures are a likely explanation for the detection of typical low-temperature combustion intermediates close to the burner surface in a low-pressure *n*-heptane flame [36]. Furthermore, the temperature gradients in the perturbed profiles are less steep compared to the unperturbed case. These observed differences between perturbed and unperturbed temperatures at these short burner-probe distances are a likely explanation for the generally observed discrepancies between experimental and modeling results close to the burner surface.

Struckmeier *et al.* showed that the centerline temperature profile, which can be approximated from the first stage pressure (or the so-called sampling function) from mass spectrometric measurements, refers to the end points of the individual temperature profiles for the different sampling locations [8]. As such, the temperature obtained is equivalent to the so-called “perturbed” profile. We have added this information to Fig. 8, and, within the given experimental uncertainties, the temperatures at the endpoints in the reaction and postflame zone can indeed be connected by the “perturbed” temperature profile obtained from molecular-beam sampling experiments. Deviations exist only for the first two millimeters and can be traced back

to uncertainties due to the steep gradients in mole fractions and temperatures that cannot be resolved by the MBMS measurements.

Furthermore, the results clearly indicate that for every sampling position, the sampled molecules experienced a different temperature history (even when radial diffusion effects are neglected). The experimental temperature profiles, measured as a function of the burner-probe distance, represent the temperatures the flame gases experience when flowing from the burner to the probe tip and as such might be important input parameters for one-dimensional model calculations. Although we have shown here that such simplified model is likely insufficient, and two-dimensional models should be developed.

Summary

We present unprecedentedly accurate, highly spatially resolved and detailed images of two-dimensional temperature fields around a quartz sampling probe in a low-pressure premixed $\text{C}_2\text{H}_4/\text{O}_2/\text{Ar}/\text{Kr}$ flame. The temperatures were determined and probe-induced perturbations revealed by Kr X-ray fluorescence from experiments that were performed at the 7-BM beamline of the Advanced Photon Source of the Argonne National Laboratory. The results clearly show how the presence of the quartz probe influences the flame temperature when sampling from the preheat, flame, and postflame region of the fuel-rich C_2H_4 flame.

Vertically, along the centerline of the burner-probe axis, temperature differences between the perturbed and the unperturbed flame of about 500 K were revealed in the postflame region. The drop in temperature becomes more significant in the reaction and preheat zones where differences between the perturbed and unperturbed flame of 1000 K have been detected. The flame temperatures are affected by the presence of the probe to about 4 mm in front of the

sampling tip. These results are generally in good agreement with previous measurements from Hartlieb *et al.* [25], Struckmeier *et al.* [8], and theoretical considerations from Gururajan *et al.* [13]. It is worth pointing out that the boundary conditions utilized for the direct numerical simulations of the non-adiabatic cases by Gururajan *et al.* [13] are actually close to what has been determined and described in this paper. In particular, they introduced heat losses to the probe surface, thus decreased the reactivity of the flame, and assumed a probe temperature of 1600 K when sampling at 6 mm above the burner surface. This is in good agreement with the current experimental results for burner-probe separations that represent “sampling” from the postflame zone.

More interestingly, the X-ray fluorescence technique allows for an accurate determination of the temperature distortions along the radial axis. A surprisingly large range of 20 mm away from the centerline appears to be affected by the presence of the sampling probe when sampling from the preheat zone. This radial distortion becomes shorter (8-10 mm) when sampling in the reaction and postflame regions of the flame.

The XRF results clearly highlight the need to develop flame models beyond the currently one-dimensional approach. Such a two-dimensional approach would eventually allow for a better description of the experimental conditions of flame-sampling experiments and therefore to more accurate insights into the ongoing combustion chemistry by providing a more realistic comparison between experimental and modeling results. In future experiments, the extent of the probe perturbations will be examined for different probe designs and flame conditions. The experiments will be designed to (a) provide more guidance towards the two-dimensional modeling approach, (b) design sampling probes that minimize the impact on the chemically detailed flame structure, and (c) develop correction rules for existing data.

Concluding Remarks

Ever since the inception of flame-sampling molecular beam mass spectrometry for flame structure analysis and the use of this chemically specific data in the form of mole fraction profiles for model validation, there has been a discussion about what temperature to use as input for model predictions. Currently the model calculations exclusively rely on the assumption of the one-dimensionality of the flame structure, thus they follow the combustion chemistry only as function of the distance to the burner surface. The presented results clearly show that the approach followed by Bastin *et al.* [23] to use the entire “perturbed” temperature profile, i.e., the temperatures measured directly in front of the nozzle, as input in the model calculations would result in an apparent shift of the temperature profile away from the burner surface and a significantly lower temperature than what the species would actually experience when flowing through the burner to the nozzle. A more accurate input for one-dimensional model calculations would be the temperature history of the sampled gases. It is now understood that samples drawn from different flame positions are equivalent to samples from different flames, due to the changes in perturbations at different sampling points. However, these effects are not implemented in current modeling approaches.

More importantly, the experimental results presented here reveal the limitations of the simple one-dimensional approach used for model calculations and interpretations of flame-sampled data. As a consequence, caution should be used when flame-sampled mole fraction profiles serve as the only validation targets for chemical kinetic models. Given the temperature distortions and the lack of one-dimensionality of the flame, a perfect agreement between experimental and one-dimensional modeling results cannot be expected. Instead, it is

recommended that two-dimensional effects should be considered when modeling the chemistry of low-pressure flame data obtained via invasive sampling techniques. Such an approach would ensure high-fidelity comparisons between experiment and model predictions.

When comparing the cost of one- vs. two dimensional flame-simulations, the computational cost needed to address two-dimensional effects might be relatively high, but the current and steadily growing computational power is expected to enable such simulations to be implemented in flame chemistry simulations to support the development of more accurate detailed chemistry mechanism. However, it is beyond the scope of the present work to show how the presented results would influence the model predictions in more sophisticated two-dimensional calculations.

Until such more sophisticated two-dimensional modeling approaches are implemented, the focus of the flame-sampling experiments should probably be more on extracting chemical insights based on the experimental data alone, e.g. species identification, rather than on the attempt to exactly match experimental mole fraction profiles with model predictions. In this context, it is suggested that comparisons of chemical structures of flames fueled by isomeric fuels under the exact same conditions might be a valid approach to obtain insights into combustion chemistry [37-49], because the influence of the probe perturbations on the results should be identical. Furthermore, an analysis of trends in signal intensities observed in single mass spectra should allow for accessing chemical information as was recently demonstrated for flame-sampled mass spectra from opposed-flow diffusion flames [50].

Despite the described probe-perturbation issues, flame-sampling techniques should still be considered indispensable for combustion chemistry studies because they allow for an extraction of comprehensive chemical information by probing a large range of chemical

structures simultaneously. It should also be pointed out that, given the combined experimental uncertainties of the invasive (probe-sampling) and non-invasive (laser-based) experimental technique, Struckmeier *et al.* [8] showed that “respectable” agreement can be achieved between invasive and non-invasive measurements of mole fractions of intermediate species.

In conclusion, for the flame-sampling data to have the highest possible impact, it seems to be advisable to reconsider the current modeling approach and to include two-dimensional effects.

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Figures

Figure 1: Schematic of the experimental setup. The z-axis is parallel to the X-ray beam, which enters from the right side. The fluorescence detector and polycapillary optic are in the foreground. The diameter of the McKenna burner is 6 cm, the shroud gas area is 1 cm wide and the dimensions of the quartz probe are shown in the Supplementary Material.

Figure 2: (a) Two-dimensional image of the Kr X-ray fluorescence from a cold gas mixture of 3850 sccm Ar and 200 sccm Kr through the McKenna-burner. The pressure was kept constant at 30 Torr and the quartz probe, as indicated in (a), was positioned 8.4 mm above the burner surface. A range from 6.5 to 9.0 mm on the y-axis and of -2.0 to 2.0 mm on the x-axis was probed. (b) Normalized Kr X-ray fluorescence signal as a function of the radial distance from the cone and (c) as a function of the distance from the burner surface.

Figure 3: (a) Two-dimensional image of the Kr X-ray fluorescence from the unperturbed $\text{C}_2\text{H}_4/\text{O}_2/\text{Ar}/\text{Kr}$ flame at a pressure of 30 Torr. A range from 23.0 to 30.0 mm on the y-axis (post flame) and of -12.0 to 0.0 mm on the x-axis was probed. (b) Temperature profiles of the unperturbed flame at various radial positions: $x = 0$ (center line), 15.0, and 25.0 mm. The temperature points obtained via the independent OH LIF measurements are shown as red stars.

Figure 4: Two-dimensional image of the Kr fluorescence signal. The image was recorded at a burner-probe separation of 4.2 mm; a range from $y = 1.5$ to 4.7 mm and from $x = -12.0$ to 0.0 mm was scanned. Darker colors indicate smaller signal intensities.

Figure 5: Two-dimensional temperature fields around the quartz probe at two different burner-probe separations: (a) 4.2 mm (reaction zone) and (b) 2.0 mm (preheat zone). Isotherms have been added to the images to guide the readers' eyes.

Figure 6: Temperature profiles as function of the radial and axial distance from the probe tip. The burner probe separation was kept at 4.2 mm. (a) X-ray burner separation 4.6 mm which places the probe tip below the X-ray beam and equates to a probe X-ray separation of -0.4 mm. (b) X-ray burner separation 4.0 mm = 0.2 mm in front of the tip. (c) X-ray burner separation 3.0 mm = 1.2 mm in front of the tip. (d) X-ray burner separation 1.5 mm = 2.7 mm in front of the probe tip.

Figure 7: Distorted temperature profiles as function of the radial distance from the cone and burner-probe separations of 2.4, 4.2, and 31.0 mm. Distance of the X-ray beam to probe tip = 0.2 mm.

Figure 8: Centerline ($x=0$) temperature profiles determined by X-ray fluorescence. The closed symbols were obtained from an unperturbed flame. The open symbols represent measurements in a perturbed flame with various burner-probe separations (2.0-31.0 mm). The burner-probe separations can be identified from the maximum HAB in a series. Vertical dotted lines indicate the deviation in temperature between the unperturbed and perturbed flames and the values are indicated above the lines. The inset panel shows the data for HAB from 16-34mm. The dashed line represents the completely "perturbed" profiles as obtained from mass spectrometric measurements following procedures described by Struckmeier *et al.* [8].